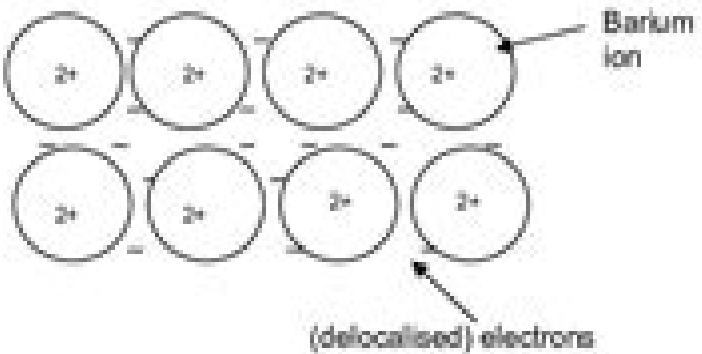


Question	Answer	Marks	Guidance
4 (a)	 <p>Diagram showing a regular arrangement of labelled '<math>\text{Ba}^{2+}</math> ions' or '<math>2^{+}</math> ions' and some attempt to show electrons ✓</p> <p>Scattering of labelled electrons between other species. <b>AND</b> statement anywhere of <b>delocalised</b> electrons (can be in text or in diagram) ✓</p> <p>The attraction between (positive) ions and (delocalised) electrons is strong ✓</p>	3	<p>Regular arrangement must have at least two rows of correctly charged ions and a minimum of two ions per row</p> <p><b>ALLOW</b> as label: positive ions, cations if correct charge is seen within circle  <b>ALLOW</b> for labelled <math>\text{Ba}^{2+}</math> ions: circles with <math>\text{Ba}^{2+}</math> inside  <b>DO NOT ALLOW</b> incorrect charge for ions eg + , <math>3^{+}</math> etc  <b>DO NOT ALLOW</b> for label of ions: nuclei <b>OR</b> positive atom <b>OR</b> protons  <b>ALLOW</b> <math>e^{-}</math> or '<math>e</math>' or <math>-</math> as symbol for electron within the lattice for first marking point if not labelled as 'electrons'.</p> <p><b>ALLOW</b> mobile or 'sea of' for delocalised</p> <p><i>Quality of written communication: 'electron(s)' spelled correctly and used in context for the <b>third</b> marking point</i>  <b>ALLOW</b> a lot of energy is needed to break <b>OR</b> overcome the attraction between (positive) ions and (delocalised) electrons  <b>IGNORE</b> 'heat' but <b>ALLOW</b> 'heat energy'  <b>DO NOT ALLOW</b> references to incorrect particles or incorrect attractions eg 'intermolecular attraction' <b>OR</b> 'nuclear attraction'</p> <p><b>IGNORE</b> 'strong metallic bonds' without seeing correct description of metallic bonding</p>

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## **Table of Contents Ocr Jan 2013 Core 4 Mark Scheme**

1. Understanding the eBook Ocr Jan 2013 Core 4 Mark Scheme
  - The Rise of Digital Reading Ocr Jan 2013 Core 4 Mark Scheme
  - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Jan 2013 Core 4 Mark Scheme
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ocr Jan 2013 Core 4 Mark Scheme
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Jan 2013 Core 4 Mark Scheme
  - Personalized Recommendations
  - Ocr Jan 2013 Core 4 Mark Scheme User Reviews and Ratings
  - Ocr Jan 2013 Core 4 Mark Scheme and Bestseller Lists
5. Accessing Ocr Jan 2013 Core 4 Mark Scheme Free and Paid eBooks
  - Ocr Jan 2013 Core 4 Mark Scheme Public Domain eBooks

- Ocr Jan 2013 Core 4 Mark Scheme eBook Subscription Services
- Ocr Jan 2013 Core 4 Mark Scheme Budget-Friendly Options
- 6. Navigating Ocr Jan 2013 Core 4 Mark Scheme eBook Formats
  - ePub, PDF, MOBI, and More
  - Ocr Jan 2013 Core 4 Mark Scheme Compatibility with Devices
  - Ocr Jan 2013 Core 4 Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ocr Jan 2013 Core 4 Mark Scheme
  - Highlighting and Note-Taking Ocr Jan 2013 Core 4 Mark Scheme
  - Interactive Elements Ocr Jan 2013 Core 4 Mark Scheme
- 8. Staying Engaged with Ocr Jan 2013 Core 4 Mark Scheme
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ocr Jan 2013 Core 4 Mark Scheme
- 9. Balancing eBooks and Physical Books Ocr Jan 2013 Core 4 Mark Scheme
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ocr Jan 2013 Core 4 Mark Scheme
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr Jan 2013 Core 4 Mark Scheme
  - Setting Reading Goals Ocr Jan 2013 Core 4 Mark Scheme
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr Jan 2013 Core 4 Mark Scheme
  - Fact-Checking eBook Content of Ocr Jan 2013 Core 4 Mark Scheme
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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