

SAMSUNG

# User Manual

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# Samsung Series 7 Manual Eject Cd

**Albert A Gayle**



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solutions of weak acids with the indicated  $pK_a$  values

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web titration curves source openstax chemistry atoms first 2e strong acid strong base titrations when titrating a strong acid such as hydrochloric acid with sodium hydroxide you are reacting the  $HCl$  directly with  $NaOH$   $HCl(aq) + NaOH(aq) \rightarrow H_2O(l) + NaCl(aq)$

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Volume of $NaOH$ added ( $mL$ )	$pH$ of the solution
4	6
4	6
5	9
5	9

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web background plot of the  $pH$  of a solution against the volume of titrant added is called a titration curve the  $pH$  can be measured directly with a  $pH$  meter while titrant is added from a buret for the acids used in this experiment the titrant will always be a  $0.1\ M$  solution of the strong base  $NaOH$

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you add another chemical that changes color at the ph equal to the equivalence point when the acid and base are in  
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