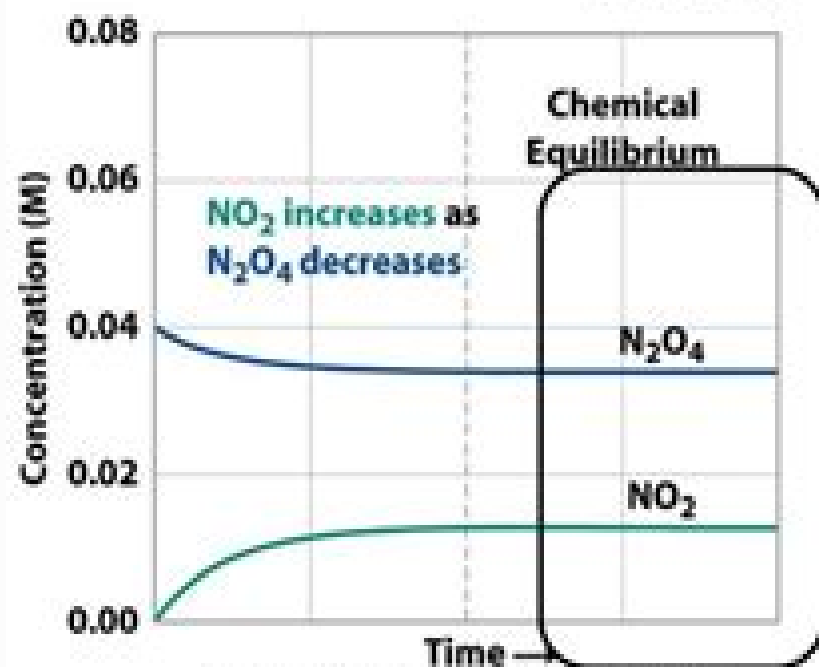
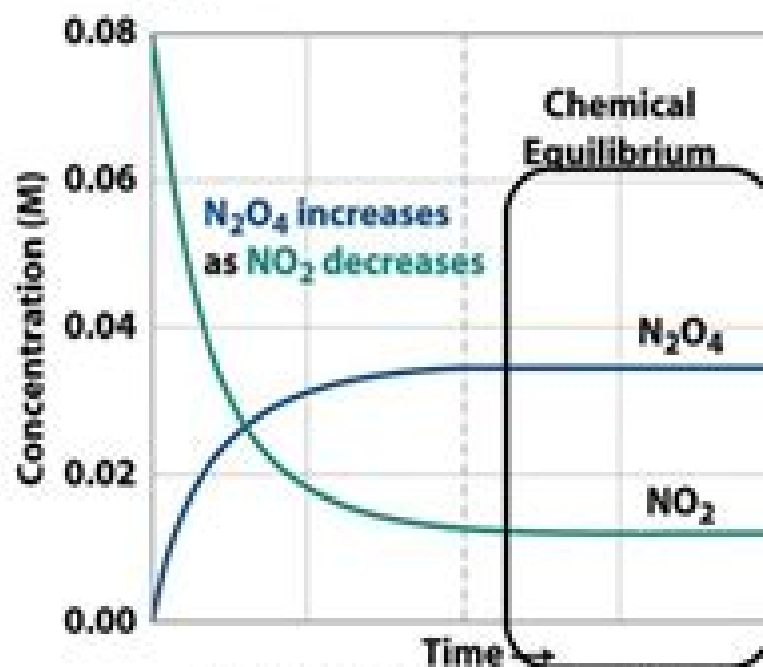


The Equilibrium State



(a) Only N_2O_4 is present initially



(b) Only NO_2 is present initially

In both experiments, a state of chemical equilibrium is reached when the concentrations level off at constant values: $[\text{N}_2\text{O}_4] = 0.0337 \text{ M}$; $[\text{NO}_2] = 0.0125 \text{ M}$.

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Thermodynamics of Geothermal Fluids Andri Stefánsson,Thomas Driesner,Pascale Bénézech,2018-12-17 Volume 76 of Reviews in Mineralogy and Geochemistry presents an extended review of the topics conveyed in a short course on Geothermal Fluid Thermodynamics held prior to the 23rd Annual V M Goldschmidt Conference in Florence Italy August 24 25 2013 It covers Thermodynamics of Geothermal Fluids The Molecular Scale Fundament of Geothermal Fluid Thermodynamics Thermodynamics of Aqueous Species at High Temperatures and Pressures Equations of State and Transport Theory Mineral Solubility and Aqueous Speciation Under Hydrothermal Conditions to 300 C The Carbonate System as an Example Thermodynamic Modeling of Fluid Rock Interaction at Mid Crustal to Upper Mantle Conditions Speciation and Transport of Metals and Metalloids in Geological Vapors Solution Calorimetry Under Hydrothermal Conditions Structure and Thermodynamics of Subduction Zone Fluids from Spectroscopic Studies and Thermodynamics of Organic Transformations in Hydrothermal Fluids

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a place for itself as the trusted resource for readers across the country With award winning writing and photography covering everything from politics and food to theater and fashion the magazine s consistent mission has been to reflect back to its audience the energy and excitement of the city itself while celebrating New York as both a place and an idea

Comparison of Numerical Techniques for Integration of Stiff Ordinary Differential Equations Arising in Combustion Chemistry Krishnan Radhakrishnan, 1984 The efficiency and accuracy of several algorithms recently developed for the efficient numerical integration of stiff ordinary differential equations are compared The methods examined include two general purpose codes EPISODE and LSODE and three codes CHEMEQ CREKID and GCKP84 developed specifically to integrate chemical kinetic rate equations The codes are applied to two test problems drawn from combustion kinetics The comparisons show that LSODE is the fastest code currently available for the integration of combustion kinetic rate equations An important finding is that an interactive solution of the algebraic energy conservation equation to compute the temperature does not result in significant errors In addition this method can be more efficient than evaluating the temperature by integrating its time derivative Significant reductions in computational work are realized by updating the rate constants $k \exp(-E/RT)$ only when the temperature change exceeds an amount ΔT that is problem dependent An approximate expression for the automatic evaluation of ΔT is derived and is shown to result in increased efficiency

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