#### Equilibrium Practice Problems

 Write the equilibrium expression for each of the following reactions:  $N_{2,(g)} + 3 H_{2,(g)} \leftrightarrow 2 NH_{3,(g)}$ 

$$K = \frac{[NH_3]^2}{[N_2][H_2]^3}$$

$$I_{2(s)} + CI_{2(g)} \leftrightarrow 2ICI_{(g)}$$

$$K = \underline{[ICI]^2}$$

$$NO_{2(q)} \leftrightarrow NO_{(q)} + V_2 O_{2(q)}$$

$$K = [NO][O_2]^{1/2}$$
  
 $[NO_2]$ 

The dissociation of acetic acid, CH<sub>3</sub>COOH, has an equilibrium constant at 25°C of 1.8 x 10°5. The reaction is

$$CH_3COOH_{(aq)} \leftrightarrow CH_3COO^-_{(aq)} + H^+_{(aq)}$$

If the equilibrium concentration of CH<sub>2</sub>COOH is 0.46 moles in 0.500 L of water and that of CH<sub>3</sub>COO is 8.1 x 10<sup>-3</sup> moles in the same 0.500 L, calculate [H+] for the reaction.

$$K = [CH_3COO+][H+]$$

 $K = [CH_3COO^*][H^*]$  Plug in known values and solve.

$$1.8 \times 10^{-5} = [8.1 \times 10^{-3} \text{ moles/0.500 L}] [H^+]$$
  
[0.46 moles/0.500 L]

$$[H^+] = 1.0 \times 10^{-3} M$$

Indicate the effect of a catalyst, pressure, temperature and concentration on each of the following on:

		catalyst	pressure	temperature	concentration
a.		presence increase	direct relationship	direct relationship	direct relationship
ь.	posn	nothing	based on moles	based on enthalpy of rxn	inverse relationship

# **Reaction Rates And Equilibrium Practice Problems**

**Xiang Xie** 

# **Reaction Rates And Equilibrium Practice Problems:**

Barron's Chemistry Practice Plus: 400+ Online Questions and Quick Study Review Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2022-07-05 Need quick review and practice to help you excel in Chemistry Barron s Chemistry Practice Plus features more than 400 online practice questions and a concise review guide that covers the basics of Chemistry Inside you ll find Concise review on the basics of Chemistry an excellent resource for students who want a quick review of the most important topics Access to 400 online questions arranged by topic for customized practice Online practice includes answer explanations with expert advice for all questions plus scoring to track your progress This essential guide is the perfect practice supplement for students and teachers **New Understanding Chemistry for Advanced Level Third** Edition Ted Lister, Janet Renshaw, 2000 Matches the specifications of the Awarding Bodies AQA NEAB AEB OCR and Edexcel This accessible text includes frequent hints questions and examination questions providing support and facilitating study at home It features photographs and comprehensive illustrations with 3D chemical structures **MCAT Practice Ouestions** & Actual Exam Dumps using AAMC format for your easy success Allied Books, The Medical College Admission Test MCAT developed and administered by the AAMC is a standardized multiple choice examination created to help medical school admissions offices assess your problem solving critical thinking and knowledge of natural behavioral and social science concepts and principles prerequisite to the study of medicine Preparing for the MCAT exam to become enter Medical College this year Here We ve brought 450 Exam Questions for you so that you can prepare well for this MCAT exam Unlike other online simulation practice tests you get an eBook version that is easy to read remember these questions You can simply rely on these questions for successfully certifying this exam Chemistry is Phenomenal Tracy Poulsen, Problems in Physical Chemistry IEE Main and Advanced Volume 1 Dr. RK Gupta, 2021-04-05 1 The book is prepared for the problem solving in chemistry 2 It is divided into 8 chapters 3 Each chapter is topically divided into guick theory Immediate Test and Knowledge Confirmation Test 4 At the end of the each chapter cumulative exercises for JEE Main Advanced for practice 5 Acid Test for JEE Mains Advance containing all types of questions asked in JEE A common phrase among JEE Aspirants that chemistry is the most scoring subject but the problems asked in JEE Exams are not directly related but they are based on multiple applications Introducing the all new edition of Problem Physical Chemistry JEE Main Advanced Volume 1 which is designed to develop the use of the concepts of chemistry in solving the diversified problems as asked in JEE The book divides the syllabus into 8 chapters and each chapter has been topically divided in guick theory different types of Solved Examination followed by Immediate Test along with the Topicwise short exercises Knowledge Confirmation Test At the end of each chapter there are separate cumulative exercises for JEE Main Advanced Acid Test for JEE Mains Advance are also provided containing all types of questions asked in JEE Detailed and explanatory solutions provided to all the questions for the better understanding TOC Mole concept and Stiochiometry Atomic Structure Stages of Matter 1 Stages of Matter 2 Thermodynamic

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