

TUTORIAL 3

ALKENES & CYCLOALKENES

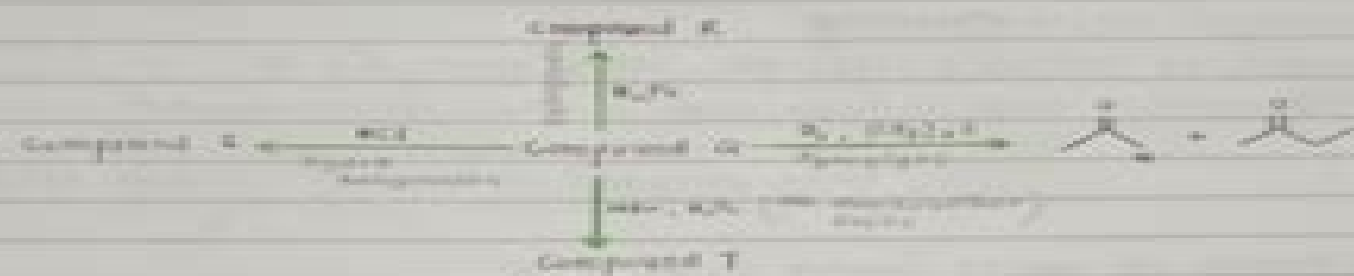
Topic

Class

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Q1

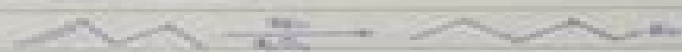
Alkene can undergo various reactions. Identify the structure of R, S, T and U.



Compound R (alkene)



Compound T (anti-Markovnikov addition)



Compound S (hydrobromide)



Compound U (hydrohalogenation)



Q2

a) Define hydrocarbon compounds

A class of organic compounds made up of only the elements carbon (C) and hydrogen (H).

b) Indicate the difference between propene and propane

Propene	Propane
- Propene is a saturated hydrocarbon compound.	- Propane is a unsaturated hydrocarbon compound.
- carbon-carbon double bond	- carbon-carbon single bond in the structure
- fully saturated to bond	- double bond and single bond

Reaction Lab Answer Key Hydrocarbon

**United States. Environmental
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Reaction Lab Answer Key Hydrocarbon:

Unveiling the Magic of Words: A Report on "**Reaction Lab Answer Key Hydrocarbon**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Reaction Lab Answer Key Hydrocarbon**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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