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followed by the name of the anion the name of the nonmetallic element with its ending replaced by the suffix ide some
examples are given in table 2 6 names of some ionic compounds table 2 6 compounds containing

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has a molar mass of one mole of the compound would result in three moles of tellurium a total of 382 8g add the mass of
bismuth and the mass of tellurium per mole to find the total molar mass divide the mass of bismuth

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these are the two types of bonds out of which every compound is made of 2 example of compounds example of compounds
includes water H_2O hydrogen peroxide H_2O_2 etc

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water are composed of molecules that is they are molecular compounds sodium chloride on the other hand contains ions it is
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bonding is an important and extensive concept in chemistry and it will be treated in considerable

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it is ionic write the symbols for the ions involved a NF_3 b BaO c NH_4Cl d SrH_2 e PO_4^{3-} f Na_2O answer a answer b
answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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and nothing else so it is pure when we mix two different pure

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name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element
with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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substances as water H_2O H_2O and carbon dioxide CO_2 CO_2 these compounds are very different from ionic compounds like
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ch 2 ch 3 answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p 4 b h 2 o c ca no 3 2 d ch 3 co 2 h acetic acid

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you need to take or know from a plan schematic three length measurements make sure they are all in the same length unit or convert accordingly until they see more

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