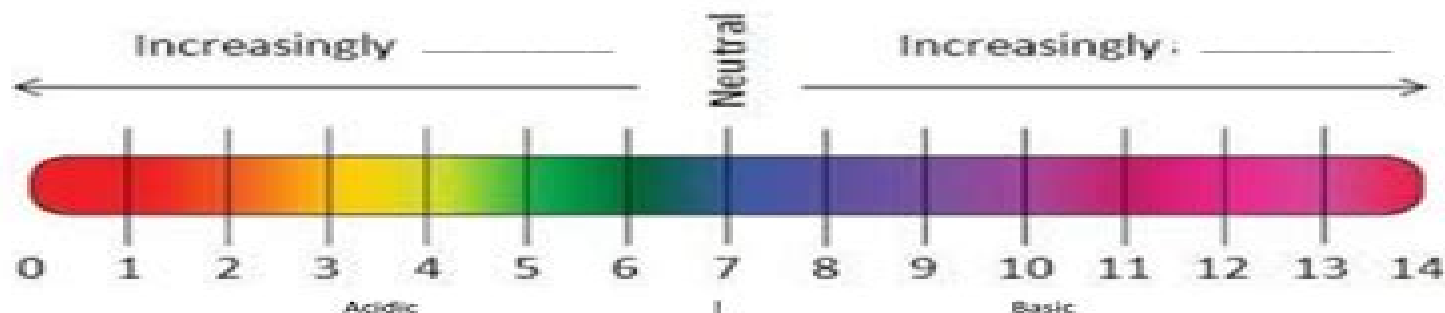


Week 7 Laboratory PhET Assignment: pH Scale

A. Explore the **pH Scale**: Select the 'Macro' tab.

1. Compare the pH scale on the left-hand side of the screen to the pH scale below. Label the one below as acidic and basic:



2. Investigate the pH of each of the substances found under the drop-down menu.
- Drag the pH sensor into the solution to see the pH reading.
 - Record the pH of the substance and whether the substance falls into the acid or base end of the pH scale.
 - To change the substance simply select from the drop-down menu.
3. Now add these substances below the pH scale on the previous page at the appropriate location based on their pH values.

Substance	pH	acid/base
Drain cleaner	13.00	Base
Hand soap	10.00	Base
Blood	7.40	Base
Spit	7.40	Base
Milk	6.50	Acid
Chicken Soup	5.80	Acid
Coffee	5.00	Acid
Orange Juice	3.50	Acid
Soda Pop	2.50	Acid
Vomit	2.00	Acid
Battery Acid	1.00	Acid

4. Using the information from the chart and simulation answer the following questions.
- What pH values correspond to acids? pH values <7.00
 - What pH values correspond to bases? pH values greater than 7.00

Phet Lab Answer Key For Ph Scale

G Orfield



Phet Lab Answer Key For Ph Scale:

Scientific and Technical Aerospace Reports ,1987 *Chemical Abstracts* ,1990 **Government Reports Announcements & Index** ,1987 **Effects of Disease on Clinical Laboratory Tests** Richard B. Friedman,Donald S. Young,1989 An aid to determine the possible cause of laboratory test abnormalities encountered in clinical practice Sections include laboratory test index disease keyword index laboratory test listings disease listings by ICD 9CM classification and references Government Reports Announcements ,1973 **Agrindex** ,1994 **Reverse Acronyms, Initialisms & Abbreviations Dictionary** Mary Rose Bonk,Pamela Dear,1999 *Science Citation Index* ,1994 Vols for 1964 have guides and journal lists *Index to Scientific Reviews* ,1987 The Compact Edition of the Oxford English Dictionary Sir James Augustus Henry Murray,1971 Micrographic reproduction of the 13 volume Oxford English dictionary published in 1933

The pH Scale Mary Griffin,2018-12-15 The pH scale measures how acidic or basic a substance is ranging from 0 to 14 Readers will learn how certain substances rank on the pH scale what happens when acids and bases are mixed and how water can make a substance either acidic or basic These significant science concepts are discussed in approachable text and supported by motivating fact boxes charts images and photographs *Ph Measurement. Specification for Ph Scale* British Standards Institute Staff,British Standards Institution,1984-12-31 pH measurement pH Range statistics Accuracy Grades quality Standard solutions Phthalates Potassium organic compounds **PH Scale** ,2006 *The PH Scale* Marie Roesser,2025-07-30 The pH scale measures how acidic or basic a substance is ranging from 0 to 14 This chemistry concept has real world applications For example strawberries grow best in soil with a specific pH Safe drinking water must have a pH within a certain range too Readers will learn more about pH the scale how substances are tested and how pH can change in this accessible volume a valuable addition to any elementary science collection Essential curricular information is supported by Learn More fact boxes labeled diagrams and a summarizing graphic organizer Electronics pH Determinations: Theory and Practice Roger G. Bates,1954 **The Meaning and Calibration of the PH Scale** ,1933 Ph Scale And Acid-base Titrations/standardization Postma Roberts Hollenberg,James M. Postma,Julian Roberts,J. Leland Hollenberg,2000-01-15

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