

ACTIVITY: Molecular Polarity Computer Simulation (PhET)

Go to: <https://phet.illinois.edu/> Click on the arrow, DOWNLOAD, OPEN and RUN the MOLECULE POLARITY program. Or google "phet molecule polarity" to find the simulation.

Introduction:

In this atomic level simulation, you will investigate how atoms' **electronegativity** values affect the bonds they produce. When two atoms form a covalent bond, a pair of electrons is shared between atoms. **Electronegativity** is a measure of a single atom's ability to attract electrons shared in that bond. In this lab you will work at your own pace, to answer a number of questions. To begin, from what you've already learned about the protons and electrons in an atom, what would cause an atom to have a **high** electronegativity value?

Why might an atom have a low electronegativity value?

Procedure:

Two Atoms

1. Turn on (check) all view options. **distill** out the molecule with your mouse.
2. Take your time and investigate how the binary compound's bond behaves when the atom's electronegativity and orientation are changed. **Do not rush through this step.** Describe the bond formed between two atoms with **similar, low** electronegativities.

Describe the bond formed between two atoms with **similar, high** electronegativities.

Describe the bond formed between two atoms with **very different** electronegativities.

Describe (in your own words) what is meant by **partial charges**, δ^- and δ^+ .

δ^- represents: _____

δ^+ represents: _____

What happens when the electric field is applied to a very polar molecule?

Why do you think this is? _____

View

- ☒ Bond Dipoles **ON**
- ☒ Partial Charges
- ☒ Bond Orientation

Surfaces

- ☐ None
- ☐ Electrostatic Potential
- ☒ Electron Density

Electric Field

- ☒ ON ☐ OFF



Phet Molecular Polarity Lab Answer Key

Nivaldo J. Tro



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New to MasteringChemistry: MasteringChemistry metadata analysis of problems/tutorials assigned in the previous edition have been used to revise end-of-chapter problems in the Third Edition. Approximately 1,000 end-of-chapter questions have been enhanced with feedback meeting instructor's need for more tutorial-like questions. Interactive versions of selected worked examples in the text have been created and are incorporated into MasteringChemistry as assignable tutorial activities, providing an office-hour-like experience. These can also be used for mobile learning through a downloadable app.

15. Pause and Predict Video Quizzes bring chemistry to life with lab demonstrations illustrating key topics in general chemistry. Students are asked to predict the outcome of experiments as they watch the videos; a set of multiple-choice questions challenges students to apply the concepts from the video to related scenarios.

8. PhET tutorials have been developed around interactive applets that foster conceptual understanding and active learning. Topics include acid-base solutions, balancing chemical equations, and molecular polarity. Multiple-choice Reading Questions are provided for each chapter, making it easy to hold students accountable for doing assigned readings before lecture. Enhanced end-of-chapter questions within MasteringChemistry, providing wrong-answer feedback, have been added. Sketch-it-type problems have been added for each chapter. Simulations cover some of the most difficult chemistry concepts and are written by the leading authors in simulation development. Select end-of-chapter questions and reading quizzes have been tagged to learning outcomes. The overall number of algorithmic and randomized problems has been increased to 40%, offering a more rounded program for departments moving to online high-stakes testing.

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