Chemistry Chapter 8 Test

(b) two electrons

____ 12. The central atom in a Lewis structure is usually the atom _____

		Choice that best completes the statement or a	mswer.	s the question.
	1.	. What type of bond would you expect in a co	impo-oun	id of oxygen and sodium?
	300	(a) ionic	C.	covalent
		b. metallic	d.	polar covalent
	2.	The symbols 5- and 5+ indicate which of th	e follo	wing?
		a. electronegativity	G.	change
		b. electrostatic force	(d)	partial charge
-	3.	Which of the following represents a nonpolar covalent molecule?		
		a. NaCl	c.	HCN
		(5) Br ₂	di.	Au
	40	. The measure of an atom's ability to attract e	dectron	s to itself in a chemical bond is called
		a. electron affinity	(E)	electronegativity
		b. polarity	d.	ionizazion energy
_	5.	. The polarity of a bond depends on the differ	ence in	the atoms'
		a. electron affinities	c.	
		(b) electronegativities	d.	A CONTRACTOR OF THE PROPERTY O
	7.00	. 87, 200 mm 21, 200 mm		N K
	150	What type of bond would you expect to fore		
		(a) ionic		polar covalent
	-	b. covalent	a.	metallic
	79,	High electronegativity differences are found		bonds.
		a. metallic		covalent
		(b) ionic metal - narmetal	d.	Pauling c
	18	To achieve an octet when combining with o	ogassinu)	m, would an atom of a Group 16 element be expected
		to lose, gain, or share electrons?		
		a. lose	K	
		(b) gain		K 19
		c. sharek		
		This cannot be determined from the give	en info	rmation, x
_	9.	Which of the following does not exist as a d	iac omnie	molecule? H N D F CI Br I
		(a) argon	· C.	chlorine
		b. hydrogen	d.	oxygen Gas M, N, S, F, Cl, S, I,
	10.	Which of these is not a covalent molecule?		
		a. F ₂ o	(E)	Lif
		b. CO e	d.	NH ₃
	11.	A dash in a Lewis structure represents		
		a. one electron	100	four electrons

d. unpaired electrons

Odysseyware Chemistry Test Answers

S Nieto

Odysseyware Chemistry Test Answers:

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