

Question			Answer	M	Guidance
6	a	i	λ is the (de Broglie) wavelength <u>associated with a particle</u> h is (a constant known as) the Planck constant mv is the momentum/mass \times speed of <u>the particle</u>	B1 B1 B1	allow object/electron for particle accept velocity for speed accept m and v separately but the particle must appear with one of the quantities to score the mark
		ii	electrons passing through a (thin) sheet of graphite are diffracted producing rings/pattern (on a fluorescent screen)	B1 A1	any suitable situation NOT Au or Al, etc. what is observed + interpretation 1 mark. Do not award the A mark alone unless a plausible situation has been suggested. Young slits type experiments etc score 0/2
		iii 1	$eV = \frac{1}{2} mv^2$ $2eV/m = v^2$ so $v = \sqrt{(2 \times 1.6 \times 10^{-19} \times 5.0 \times 10^5 / 9.1 \times 10^{-31})}$ so $v = 1.3 \times 10^8 \text{ (m s}^{-1}\text{)}$	C1 C1 B1 A0	allow $eV = 8.0 \times 10^{-14} \text{ J}$ for 1 mark allow $\sqrt{(2 \times 8.0 \times 10^{-14} / 9.1 \times 10^{-31})}$ or evidence of correct calculation, e.g. $v = 1.325 \times 10^8$
		iii 2	$\lambda = h/mv = 6.63 \times 10^{-34} / 1.3 \times 10^8 \times 9.1 \times 10^{-31}$ $\lambda = 5.6 \times 10^{-12} \text{ (m)}$	C1 A1	allow 5.5×10^{-12} if 1.325×10^8 is used
		iii 3	λ of visible light $5.6 \times 10^{-7} \text{ m}$ so power of 10 = 5 or -5	B1 B1	ect all 2; accept 4 to $7 \times 10^{-7} \text{ m}$ NOT 10^{-7} NOT 10^5 on the answer line
	b		The photoelectric effect	B1	QWC mark allow exp't description: uv light shone on a zinc plate connected to a gle max 3 from 6 marking points
			individual photons are absorbed by individual electrons in the metal surface, i.e. one to one interaction.	B1	
			Only photons with energies above the work function energy will cause photoelectron emission/idea of threshold frequency	B1	
			Hence u-v photons or blue photons will cause photoemission but red photons will not.	B1	
			Number of electrons emitted depends on light intensity	B1	
			A wave model does not explain instantaneous emission of electrons.	B1	
			A wave model does not explain a threshold frequency/wavelength for emission to occur	B1	allow wave model does not explain no emission however bright the light if energy of photon below work function
			Total question 6	16	

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Table of Contents Ocr Physics G482 June 2013 Mark Scheme

1. Understanding the eBook Ocr Physics G482 June 2013 Mark Scheme
 - The Rise of Digital Reading Ocr Physics G482 June 2013 Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Physics G482 June 2013 Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Physics G482 June 2013 Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Physics G482 June 2013 Mark Scheme
 - Personalized Recommendations
 - Ocr Physics G482 June 2013 Mark Scheme User Reviews and Ratings
 - Ocr Physics G482 June 2013 Mark Scheme and Bestseller Lists
5. Accessing Ocr Physics G482 June 2013 Mark Scheme Free and Paid eBooks
 - Ocr Physics G482 June 2013 Mark Scheme Public Domain eBooks
 - Ocr Physics G482 June 2013 Mark Scheme eBook Subscription Services
 - Ocr Physics G482 June 2013 Mark Scheme Budget-Friendly Options

6. Navigating Ocr Physics G482 June 2013 Mark Scheme eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr Physics G482 June 2013 Mark Scheme Compatibility with Devices
 - Ocr Physics G482 June 2013 Mark Scheme Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Physics G482 June 2013 Mark Scheme
 - Highlighting and Note-Taking Ocr Physics G482 June 2013 Mark Scheme
 - Interactive Elements Ocr Physics G482 June 2013 Mark Scheme
8. Staying Engaged with Ocr Physics G482 June 2013 Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Physics G482 June 2013 Mark Scheme
9. Balancing eBooks and Physical Books Ocr Physics G482 June 2013 Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Physics G482 June 2013 Mark Scheme
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr Physics G482 June 2013 Mark Scheme
 - Setting Reading Goals Ocr Physics G482 June 2013 Mark Scheme
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr Physics G482 June 2013 Mark Scheme
 - Fact-Checking eBook Content of Ocr Physics G482 June 2013 Mark Scheme
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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