

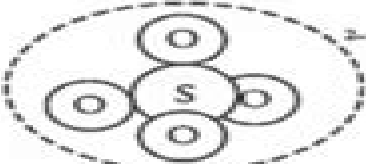
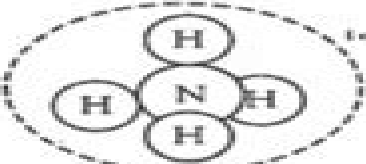
Polyatomic Ions

Can a group of atoms have a charge?

Why?

Do you know you eat a lot of "-ates"? Next time you look at a food label, read the ingredients and you will likely find a number of ingredients that end with "-ate," such as sodium phosphate or calcium carbonate. Did you ever wonder what the chemical formulas of these ingredients look like? In this activity we will explore polyatomic ions, which are groups of atoms that carry a charge. These ions are found in our food ingredients, natural waterways, and many other chemical compounds you encounter every day.

Model 1 – Types of Ions

Monatomic Ions	Nitride 	Sulfide 	Chloride 
Polyatomic Ions	Nitrate 	Sulfate 	Ammonium 
	Nitrite 	Sulfite 	Hydroxide 

1. Use Model 1 to complete the table below.

Name of Ion	Nitride	Nitrate	Sulfate	Sulfite	Ammonium
Charge on Ion	-3	-1	-2	-2	+1
Type and Number of Atoms	1 nitrogen	1 nitrogen 3 oxygen	1 sulfur 4 oxygen	1 sulfur 3 oxygen	1 nitrogen 4 hydrogen
Chemical Formula	N^{3-}	NO_3^{1-}	SO_4^{2-}	SO_3^{2-}	NH_4^{1+}

Polyatomic Ions Answer Key

Victor M. Corman

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Polyatomic Ions Answer Key

1. Understanding the eBook Polyatomic Ions Answer Key
 - The Rise of Digital Reading Polyatomic Ions Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ions Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Polyatomic Ions Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ions Answer Key
 - Personalized Recommendations
 - Polyatomic Ions Answer Key User Reviews and Ratings
 - Polyatomic Ions Answer Key and Bestseller Lists
5. Accessing Polyatomic Ions Answer Key Free and Paid eBooks
 - Polyatomic Ions Answer Key Public Domain eBooks
 - Polyatomic Ions Answer Key eBook Subscription Services
 - Polyatomic Ions Answer Key Budget-Friendly Options

6. Navigating Polyatomic Ions Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Polyatomic Ions Answer Key Compatibility with Devices
 - Polyatomic Ions Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polyatomic Ions Answer Key
 - Highlighting and Note-Taking Polyatomic Ions Answer Key
 - Interactive Elements Polyatomic Ions Answer Key
8. Staying Engaged with Polyatomic Ions Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polyatomic Ions Answer Key
9. Balancing eBooks and Physical Books Polyatomic Ions Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polyatomic Ions Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Polyatomic Ions Answer Key
 - Setting Reading Goals Polyatomic Ions Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Polyatomic Ions Answer Key
 - Fact-Checking eBook Content of Polyatomic Ions Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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