



Reinforcement

Chemical Changes

Read the passage and answer the following questions:



1. What are the products of this chemical reaction?

As mentioned in the introduction, the burning of methane in this chemical reaction, the products are water and carbon dioxide. The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

2. What is the reactant in this chemical reaction? Is it a solid, liquid, gas, or plasma? Is it a pure substance or a mixture?

3. What is the physical state of the reactant and the product?

The reactant, hydrogen peroxide, is a colorless liquid. The products, water and oxygen, are both colorless gases. The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

4. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much water and oxygen will be produced?

100 grams



5. What is the balanced chemical equation for this reaction? Is it?

6. What is the balanced chemical equation for this reaction? Is it?

7. What are the physical states of the reactant and the product?

Reactant: liquid
Product: gas

8. What is the physical state of the reactant and the product?

9. What is the balanced chemical equation for this reaction? Is it?

The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

10. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much water and oxygen will be produced?

Section 3 Reinforcement Physical And Chemical Changes Answers

Sebastian Brünink



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