# Chapter 9

# Section 3 Limiting Reactants and Percentage Yield

# Percentage Yield

- The theoretical yield is the maximum amount of product that can be produced from a given amount of reactant.
- The actual yield of a product is the measured amount of that product obtained from a reaction.
- The percentage yield is the ratio of the actual yield to the theoretical yield, multiplied by 100.

percentage yield =  $\frac{\text{actual yield}}{\text{theorectical yield}} \times 100$ 



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# **Section Quiz Limiting Reactants And Percentage Yield**

# **NARAYAN CHANGDER**

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