

Chapter 4 Atomic Structure

Section 4.3 Modern Atomic Theory**(pages 113–118)***This section focuses on the arrangement and behavior of electrons in atoms.***Reading Strategy (page 113)**

Sequencing After you read, complete the description in the flow chart below of how the gain or loss of energy affects electrons in atoms. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Bohr's Model of the Atom (pages 113–116)**

1. Circle the letter of the sentence that tells how Bohr's model of the atom differed from Rutherford's model.
 - a. Bohr's model focused on the nucleus.
 - b. Bohr's model focused on the protons.
 - c. Bohr's model focused on the neutrons.
 - d. Bohr's model focused on the electrons.
2. Is the following sentence true or false? In Bohr's model of the atom, electrons have a constant speed and move in fixed orbits around the nucleus. _____
3. What can happen to an electron in an atom when the atom gains or loses energy? _____

4. What evidence do scientists have that electrons can move from one energy level to another? _____

5. Is the following sentence true or false? When electrons release energy, some of the energy may be released as visible light. _____

Electron Cloud Model (page 116)

6. Is the following sentence true or false? Bohr's model was correct in assigning energy levels to electrons. _____
7. When trying to predict the locations and motions of electrons in atoms, scientists must work with _____.
8. What is an electron cloud? _____

Section 4 3 Modern Atomic Theory Quiz

Michael Hoelscher



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Table of Contents Section 4 3 Modern Atomic Theory Quiz

1. Understanding the eBook Section 4 3 Modern Atomic Theory Quiz
 - The Rise of Digital Reading Section 4 3 Modern Atomic Theory Quiz
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 4 3 Modern Atomic Theory Quiz
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 4 3 Modern Atomic Theory Quiz
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 4 3 Modern Atomic Theory Quiz
 - Personalized Recommendations

- Section 4 3 Modern Atomic Theory Quiz User Reviews and Ratings
- Section 4 3 Modern Atomic Theory Quiz and Bestseller Lists
- 5. Accessing Section 4 3 Modern Atomic Theory Quiz Free and Paid eBooks
 - Section 4 3 Modern Atomic Theory Quiz Public Domain eBooks
 - Section 4 3 Modern Atomic Theory Quiz eBook Subscription Services
 - Section 4 3 Modern Atomic Theory Quiz Budget-Friendly Options
- 6. Navigating Section 4 3 Modern Atomic Theory Quiz eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 4 3 Modern Atomic Theory Quiz Compatibility with Devices
 - Section 4 3 Modern Atomic Theory Quiz Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 4 3 Modern Atomic Theory Quiz
 - Highlighting and Note-Taking Section 4 3 Modern Atomic Theory Quiz
 - Interactive Elements Section 4 3 Modern Atomic Theory Quiz
- 8. Staying Engaged with Section 4 3 Modern Atomic Theory Quiz
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 4 3 Modern Atomic Theory Quiz
- 9. Balancing eBooks and Physical Books Section 4 3 Modern Atomic Theory Quiz
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 4 3 Modern Atomic Theory Quiz
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 4 3 Modern Atomic Theory Quiz
 - Setting Reading Goals Section 4 3 Modern Atomic Theory Quiz
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 4 3 Modern Atomic Theory Quiz
 - Fact-Checking eBook Content of Section 4 3 Modern Atomic Theory Quiz

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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