

Name: _____

The Organization of the Periodic Table – Answer Key

Directions

Using your computer, tablet or mobile device, navigate to the website: www.ptable.com
You will interact with this periodic table to find the answers to the questions below.

*Make sure the "Wikipedia" tab is selected in order to answer the following questions:

1. The vertical columns on the periodic table are called *groups*, how many groups are there on the periodic table? **18**
2. The horizontal rows on the periodic table are called *periods*, how many periods are there on the periodic table? **7**
3. Using the group and period numbers, identify the elements that are located in each of the following location.
 - a. The element in group 10 and period 5 = **Pd**
 - b. The element in group 15 and period 4 = **As**
 - c. The element in group 2 and period 3 = **Mg**
 - d. The element in group 18 and period 6 = **Rn**
 - e. The element in group 1 and period 7 = **Fr**
4. Are most of the elements on the periodic table classified as a metal or a non-metal?
Metal
5. Are the non-metal elements located on the left or right side of the periodic table?
Right (except for Hydrogen)

*Elements can also be classified by *family*, use the color coding at the top of the page to help identify the family names and their element members.

6. How many elements belong to the Alkali Metal family? List the member elements by their symbol.
6 elements = Li, Na, K, Rb, Cs, Fr
7. Click on the group number for the Alkali Metals family at the top of the table. Answer the following about the Alkali metals:
 - a. What are 3 similar properties of these elements? **Shiny, soft, highly reactive**
 - b. Where are these found naturally? **Only in salts, not as a free element**
 - c. What is a common substance that alkali metals react vigorously with? **Water**

Section 3 The Periodic Table Answer Key

Judd E. Hollander



Section 3 The Periodic Table Answer Key:

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the answers To find the answers a scientist applies the process skills taught in this book This program will teach you to plan work think and communicate like a scientist Back cover *Understanding Earth Student Study Guide* Peter L. Kresan, Reed Mencke, 2006-05-03 The guide helps students prepare for lectures and exams with a heavy emphasis on utilizing the book's Web resources General Chemistry Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value **The Practice of Chemistry** Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POC/PowerPoint_Final.ppt 832KB

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