

# Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 2

Q1.	What is	the	mass	of one	atom	of C -	12 in gr	ams?	

a.) 1.992 × 10<sup>-23</sup> g b.) 1.989 × 10<sup>-23</sup> g c.) 1.892 × 10<sup>-23</sup> g d.) 1.965 × 10<sup>-23</sup>

Q2. Which of the following laws states that the total mass of the reactants is equal to the total mass of the products in a chemical reaction?

a.) Law of multiple proportions

b.) Law of conservation of mass

c.) Avogadro's law

d.) Law of constant proportions

Q3. The law of multiple proportions was stated by:

a.) Dalton

b.) Proust

c.) Avogadro

d.) None of the above

Q4. The units kg m<sup>-1</sup> s<sup>-2</sup> are of \_\_\_\_

a.) Force

b.) Pressure

c.) Acceleration

d.) Velocity

Q5. The empirical formula of glucose is \_\_\_\_\_.

a.) CH<sub>2</sub>O

b.) CHO<sub>2</sub>

c.) C<sub>2</sub>H<sub>2</sub>O

d.) CH<sub>2</sub>O<sub>2</sub>

Q6. Express the following in the scientific notation:

i.) 0.0048

# **Self Check Reading And Study Workksheet Chemistry Chapter 1**

**Robert R. Redfield** 

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