

Balancing Equations Worksheet II

Count the number of atoms, then use coefficients to balance each equation.

- 1) $\underline{\hspace{1cm}} \text{AlBr}_3 + \underline{\hspace{1cm}} \text{K} \rightarrow \underline{\hspace{1cm}} \text{KBr} + \underline{\hspace{1cm}} \text{Al}$
- 2) $\underline{\hspace{1cm}} \text{FeO} + \underline{\hspace{1cm}} \text{PdF}_2 \rightarrow \underline{\hspace{1cm}} \text{FeF}_2 + \underline{\hspace{1cm}} \text{PdO}$
- 3) $\underline{\hspace{1cm}} \text{P}_4 + \underline{\hspace{1cm}} \text{Br}_2 \rightarrow \underline{\hspace{1cm}} \text{PBr}_3$
- 4) $\underline{\hspace{1cm}} \text{LiCl} + \underline{\hspace{1cm}} \text{Br}_2 \rightarrow \underline{\hspace{1cm}} \text{LiBr} + \underline{\hspace{1cm}} \text{Cl}_2$
- 5) $\underline{\hspace{1cm}} \text{PbBr}_2 + \underline{\hspace{1cm}} \text{HCl} \rightarrow \underline{\hspace{1cm}} \text{HBr} + \underline{\hspace{1cm}} \text{PbCl}_2$
- 6) $\underline{\hspace{1cm}} \text{CoBr}_3 + \underline{\hspace{1cm}} \text{CaSO}_4 \rightarrow \underline{\hspace{1cm}} \text{CaBr}_2 + \underline{\hspace{1cm}} \text{Co}_2(\text{SO}_4)_3$
- 7) $\underline{\hspace{1cm}} \text{Na}_3\text{P} + \underline{\hspace{1cm}} \text{CaF}_2 \rightarrow \underline{\hspace{1cm}} \text{NaF} + \underline{\hspace{1cm}} \text{Ca}_3\text{P}_2$
- 8) $\underline{\hspace{1cm}} \text{Mn} + \underline{\hspace{1cm}} \text{HI} \rightarrow \underline{\hspace{1cm}} \text{H}_2 + \underline{\hspace{1cm}} \text{MnI}_3$
- 9) $\underline{\hspace{1cm}} \text{Li}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{NaBr} \rightarrow \underline{\hspace{1cm}} \text{Na}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{LiBr}$
- 10) $\underline{\hspace{1cm}} \text{CaF}_2 + \underline{\hspace{1cm}} \text{Li}_2\text{SO}_4 \rightarrow \underline{\hspace{1cm}} \text{CaSO}_4 + \underline{\hspace{1cm}} \text{LiF}$
- 11) $\underline{\hspace{1cm}} \text{HBr} + \underline{\hspace{1cm}} \text{Mg}(\text{OH})_2 \rightarrow \underline{\hspace{1cm}} \text{MgBr}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 12) $\underline{\hspace{1cm}} \text{LiNO}_3 + \underline{\hspace{1cm}} \text{CaBr}_2 \rightarrow \underline{\hspace{1cm}} \text{Ca}(\text{NO}_3)_2 + \underline{\hspace{1cm}} \text{LiBr}$
- 13) $\underline{\hspace{1cm}} \text{AgNO}_3 + \underline{\hspace{1cm}} \text{Li} \rightarrow \underline{\hspace{1cm}} \text{LiNO}_3 + \underline{\hspace{1cm}} \text{Ag}$
- 14) $\underline{\hspace{1cm}} \text{Si}(\text{OH})_4 + \underline{\hspace{1cm}} \text{NaBr} \rightarrow \underline{\hspace{1cm}} \text{SiBr}_4 + \underline{\hspace{1cm}} \text{NaOH}$
- 15) $\underline{\hspace{1cm}} \text{NaCN} + \underline{\hspace{1cm}} \text{CuCO}_3 \rightarrow \underline{\hspace{1cm}} \text{Na}_2\text{CO}_3 + \underline{\hspace{1cm}} \text{Cu}(\text{CN})_2$

Challenge: $\underline{\hspace{1cm}} \text{C}_2\text{H}_6 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$

Section Reactions And Equations Worksheet Answers

L Reisser



Section Reactions And Equations Worksheet Answers:

Reviewing **Section Reactions And Equations Worksheet Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Section Reactions And Equations Worksheet Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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