

# DESCRIBING CHEMICAL REACTIONS

## Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation

## Vocabulary

- chemical equation
- skeleton equation
- catalyst
- coefficients
- balanced equation

## Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A chemical reaction can be concisely represented by a chemical \_\_\_\_1\_\_\_\_. The substances that undergo a chemical change are the \_\_\_\_2\_\_\_\_. The new substances formed in a chemical reaction are the \_\_\_\_3\_\_\_\_. In accordance with the law of conservation of \_\_\_\_4\_\_\_\_, a chemical equation must be balanced. When balancing an equation, you place \_\_\_\_5\_\_\_\_ in front of reactants and products so that the same number of atoms of each \_\_\_\_6\_\_\_\_ are on each side of the equation. An equation must never be balanced by changing the \_\_\_\_7\_\_\_\_ in the chemical formula of a substance.

Special symbols are used to show the physical state of a substance in a reaction. The symbol for a liquid is \_\_\_\_8\_\_\_\_; for a solid, \_\_\_\_9\_\_\_\_; and for a gas, \_\_\_\_10\_\_\_\_. A substance dissolved in water is designated \_\_\_\_11\_\_\_\_. If a \_\_\_\_12\_\_\_\_ is used to increase the rate of a chemical reaction, its formula is written above the arrow.

1. \_\_\_\_\_
2. \_\_\_\_\_
3. \_\_\_\_\_
4. \_\_\_\_\_
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11. \_\_\_\_\_
12. \_\_\_\_\_

## Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- \_\_\_\_\_ 13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.
- \_\_\_\_\_ 14. The symbol  $\Delta$  placed over the arrow in an equation means that heat is supplied to the reaction.
- \_\_\_\_\_ 15. Atoms are destroyed in a chemical reaction.
- \_\_\_\_\_ 16. A skeleton equation is not a balanced equation.

## Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions. I
- a. Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.
  - b. Acetylene gas ( $C_2H_2$ ) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

# Section Quiz Describing Chemical Reactions Answers

**Ralph Thomas Myers**



## **Section Quiz Describing Chemical Reactions Answers:**

## **Section Quiz Describing Chemical Reactions Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Section Quiz Describing Chemical Reactions Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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