

13.1**THE NATURE OF GASES****Section Review****Objectives**

- Describe the assumptions of the kinetic theory as it applies to gases
- Interpret gas pressure in terms of kinetic theory
- Define the relationship between Kelvin temperature and average kinetic energy

Vocabulary

- | | | |
|------------------|------------------------|-----------------------------|
| • kinetic energy | • vacuum | • pascal (Pa) |
| • kinetic theory | • atmospheric pressure | • standard atmosphere (atm) |
| • gas pressure | • barometer | |

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

The kinetic theory describes the _____ of particles in matter and the forces of attraction between them. The theory assumes that the volume occupied by a gas is mostly _____, that the particles of gas are relatively _____, move _____ of each other, and are in constant _____ motion. The _____ between particles are perfectly elastic so that the total _____ remains constant. Gas pressure results from the simultaneous collisions of billions of particles with an object. Barometers are used to measure _____ pressure. Standard conditions are defined as a temperature of _____ and a pressure of _____.

1. _____
2. _____
3. _____ **10** _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

8
9

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 11. Atmospheric pressure is 760 mm Hg.
- _____ 12. The SI unit of pressure is the pascal.

Section Characteristics Of Gases Concept Review

Tobias Bleicker



Section Characteristics Of Gases Concept Review:

Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Section Characterisrtics Of Gases Concept Review**

In a global inundated with screens and the cacophony of immediate communication, the profound energy and psychological resonance of verbal art usually diminish in to obscurity, eclipsed by the constant assault of sound and distractions. However, nestled within the lyrical pages of **Section Characterisrtics Of Gases Concept Review**, a fascinating function of fictional brilliance that pulses with natural thoughts, lies an memorable journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, this exciting opus books readers on a mental odyssey, softly exposing the latent potential and profound influence embedded within the complex internet of language. Within the heart-wrenching expanse of the evocative evaluation, we will embark upon an introspective exploration of the book is central themes, dissect their captivating publishing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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