

Common Name	Chemical Name	Formula
baking soda	sodium hydrogen carbonate	NaHCO_3
bleach (liquid)	sodium bicarbonate sodium hypochlorite or	NaClO
bleach (solid)	hydrogen peroxide	H_2O_2
Borax	sodium perborate	NaBO_3
brimstone	sodium tetraborate	$\text{Na}_2\text{B}_4\text{O}_7 \cdot 10 \text{H}_2\text{O}$
cream of tartar	decahydrate sulfur	S
Epsom salt	potassium hydrogen tartrate	$\text{KHC}_4\text{H}_4\text{O}_6$
Freon	magnesium sulfate heptahydrate	$\text{MgSO}_4 \cdot 7 \text{H}_2\text{O}$
galena	dichlorodifluoromethane	CF_2Cl_2
grain alcohol	lead (II) sulfide	PbS
graphite	ethanol	$\text{C}_2\text{H}_5\text{OH}$
gypsum	carbon	C
hypo	calcium sulfate dihydrate	$\text{CaSO}_4 \cdot 2 \text{H}_2\text{O}$
laughing gas	sodium thiosulfate	$\text{Na}_2\text{S}_2\text{O}_3$
lime	dinitrogen oxide	N_2O
limestone	calcium oxide	CaO
lye	calcium carbonate	CaCO_3
marble	sodium hydroxide	NaOH
MEK	calcium carbonate	CaCO_3
milk of magnesia	ethyl methyl ketone	$\text{CH}_3\text{COC}_2\text{H}_5$
muriatic acid	magnesium hydroxide	Mg(OH)_2
oil of vitriol	hydrochloric acid	HCl
plaster of paris	sulfuric acid	H_2SO_4
potash	calcium sulfate $\blacklozenge$ hydrate	$\text{CaSO}_4 \cdot \blacklozenge \text{H}_2\text{O}$
pyrite (fool $\blacklozenge$ s gold)	potassium carbonate	K_2CO_3
quartz	iron disulfide	FeS_2
quicksilver	silicon dioxide	SiO_2
rubbing alcohol	mercury	Hg
sal ammoniac	isopropyl alcohol	$(\text{CH}_3)_2\text{CHOH}$
salt	ammonium chloride	NH_4Cl
salt substitute	sodium chloride	NaCl
saltpeter	potassium chloride	KCl
slaked lime	potassium nitrate	KNO_3
sugar	calcium hydroxide	Ca(OH)_2
trisodium phosphate	sucrose	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$
washing soda	sodium phosphate	Na_3PO_4
wood alcohol	sodium carbonate decahydrate	$\text{Na}_2\text{CO}_3 \cdot 10 \text{H}_2\text{O}$
	methyl alcohol	CH_3OH

Section 6 Chemical Names And Formulas Key

LM Reder



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Congressional Record United States. Congress,1966 **The Journal of Educational Research** ,1926 **Industrial & Engineering Chemistry** ,1923 Chemistry Workbook For Dummies Peter J. Mikulecky,Katherine Brutlag,Michelle Rose Gilman,Brian Peterson,2008-08-06 From liquids and solids to acids and bases work chemistry equations and use formulas with ease Got a grasp on the chemistry terms and concepts you need to know but get lost halfway through a problem or worse yet not know where to begin Have no fear this hands on guide helps you solve many types of chemistry problems in a focused step by step manner With problem solving shortcuts and lots of practice exercises you ll build your chemistry skills and improve your performance both in and out of the science lab You ll see how to work with numbers atoms and elements make and remake compounds understand changes in terms of energy make sense of organic chemistry and more 100s of Problems Know where to begin and how to solve the most common chemistry problems Step by step answer sets clearly identify where you went wrong or right with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence **Fundamentals of Chemistry** Elizabeth P. Rogers,1987

Pulp and Paper Magazine of Canada ,1925 *Questions Set at the Examinations* ,1924 *Industrial and Engineering Chemistry* ,1925 **Handbook of Natural Pesticides** E. David Morgan,N. Bhushan Mandava,1988-02-29 Overzicht van diverse natuurlijk voorkomende chemische stoffen met biologische activiteit Deze stoffen worden gewonnen uit planten insekten en diverse micro organismen Aandacht voor ziektebestrijdingsmaatregelen allelopathie en allelochemicalien stoffen die een natuurlijke bescherming geven chemische boodschappers en insektengedrag afhankelijkheid door insekten van sensorische stoffen in verband met de reproductie wetgeving en registratievoorschriften van pesticiden biotoetsen voor plantehormonen andere natuurlijk voorkomende groeiregulatoren insekten en insektenpathogenen *Questions Set at the Comprehensive Examinations of ...* ,1924 **Fundamentals of Sustainable Chemical Science** Stanley E.

Manahan, 2009-03-10 Written by Stanley Manahan Fundamentals of Sustainable Chemical Science has been carefully designed to provide a basic introduction to chemistry including organic chemistry and biochemistry for readers with little or no prior background in the subject Manahan bestselling author of many environmental texts presents the material in a practical

Federal Register, 1968-04 *Decorator*, 1925

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Section 6 Chemical Names And Formulas Key**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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