

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte/100% Gesamtpunkte)

Die nachstehenden Probleme betreffen die Veranschaulichung des Verhaltens der Stoffeigenschaften mittels "Diagramme" von Molanteile, Druck, Temperatur in der Zustandsform. Bei diesen Stoffen werden Stoffeigenschaften mittels 22.04.2020

$$M(P_1, T_1) = M(P_2, T_2)$$

12. Wasserstoff (molekulare Masse $M = 2$) wird in einem mit 10 MPa (Druck) und bei einer bestimmten Temperatur gefüllten Zylinder (Volumen $V = 1$) bei 20 MPa (Druck) (10 MPa) gehalten.

$$T_1 = 200 \text{ K} \quad T_2 = 200 \text{ K} \quad T_3 = 200 \text{ K} \quad T_4 = 200 \text{ K}$$

13. Wasserstoff (molekulare Masse $M = 2$) wird in einem mit 10 MPa (Druck) und bei einer bestimmten Temperatur gefüllten Zylinder (Volumen $V = 1$) bei 20 MPa (Druck) (10 MPa) gehalten.

$$T_1 = 200 \text{ K} \quad T_2 = 200 \text{ K} \quad T_3 = 200 \text{ K} \quad T_4 = 200 \text{ K}$$

14. Wasserstoff (molekulare Masse $M = 2$) wird in einem mit 10 MPa (Druck) und bei einer bestimmten Temperatur gefüllten Zylinder (Volumen $V = 1$) bei 20 MPa (Druck) (10 MPa) gehalten.

$$T_1 = 200 \text{ K} \quad T_2 = 200 \text{ K} \quad T_3 = 200 \text{ K} \quad T_4 = 200 \text{ K}$$

15. In einem Zylinder (Volumen $V = 1$) wird Wasser (molekulare Masse $M = 18$) bei einer bestimmten Temperatur (Druck $P = 10$ MPa) gehalten. Die Temperatur wird auf 20 MPa (Druck) (10 MPa) erhöht. Die Temperatur wird auf 20 MPa (Druck) (10 MPa) erhöht. Die Temperatur wird auf 20 MPa (Druck) (10 MPa) erhöht.

$$T_1 = 200 \text{ K} \quad T_2 = 200 \text{ K} \quad T_3 = 200 \text{ K} \quad T_4 = 200 \text{ K}$$

16. Wasserstoff (molekulare Masse $M = 2$) wird in einem mit 10 MPa (Druck) und bei einer bestimmten Temperatur gefüllten Zylinder (Volumen $V = 1$) bei 20 MPa (Druck) (10 MPa) gehalten.

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DJ Losen



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Table of Contents Section 3 Gases Ch 1review Answers

1. Understanding the eBook Section 3 Gases Ch 1review Answers
 - The Rise of Digital Reading Section 3 Gases Ch 1review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 3 Gases Ch 1review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 3 Gases Ch 1review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 3 Gases Ch 1review Answers
 - Personalized Recommendations
 - Section 3 Gases Ch 1review Answers User Reviews and Ratings
 - Section 3 Gases Ch 1review Answers and Bestseller Lists
5. Accessing Section 3 Gases Ch 1review Answers Free and Paid eBooks

- Section 3 Gases Ch 1review Answers Public Domain eBooks
- Section 3 Gases Ch 1review Answers eBook Subscription Services
- Section 3 Gases Ch 1review Answers Budget-Friendly Options
- 6. Navigating Section 3 Gases Ch 1review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 3 Gases Ch 1review Answers Compatibility with Devices
 - Section 3 Gases Ch 1review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 3 Gases Ch 1review Answers
 - Highlighting and Note-Taking Section 3 Gases Ch 1review Answers
 - Interactive Elements Section 3 Gases Ch 1review Answers
- 8. Staying Engaged with Section 3 Gases Ch 1review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 3 Gases Ch 1review Answers
- 9. Balancing eBooks and Physical Books Section 3 Gases Ch 1review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 3 Gases Ch 1review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 3 Gases Ch 1review Answers
 - Setting Reading Goals Section 3 Gases Ch 1review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 3 Gases Ch 1review Answers
 - Fact-Checking eBook Content of Section 3 Gases Ch 1review Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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