

3 Gas Laws

Concept Review Questions

1. If you made a barometer and filled it with a liquid one third as dense as mercury, how high would the level of the liquid be on a day of normal atmospheric pressure?
2. What happens when the pressure acting on a gas is held constant but the temperature of the gas is changed?
3. What happens when the temperature of a gas remains constant and pressure is changed?
4. If a balloon filled with air is at rest, then the average velocity of the particles is zero. Does that mean the assumptions of the kinetic theory are not good? Explain.

Reviewing Concepts

1. What is the source of atmospheric pressure?
2. State standard atmospheric pressure in four different units.
3. Why does the air pressure in automobile tires increase when the car is driven on a hot day?
4. An air cylinder with a movable piston is placed in a refrigerator. The volume shrinks. How could you increase the volume to its original size without removing the piston from the refrigerator?
5. Describe how a real gas differs from an ideal gas. What are some of the consequences of these differences?

Section The Gas Laws Concept Review

A Gutmann



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