

1. Valence Electrons & Chemical Bonds

Table 7.1

Electron-Dot Structures

Group	1	2	13	14	15	16	17	18
Diagram	Li [•]	Be ^{••}	B ^{•••}	C ^{••••}	N ^{•••••}	O ^{••••••}	F ^{•••••••}	Ne ^{••••••••}

- A chemical bond is the force that holds two atoms together.
- Chemical bonds form by the attraction between the positive nucleus of one atom and the negative electrons of another atom.
- Atom's try to form the octet—the stable arrangement of eight valence electrons in the outer energy level—by gaining or losing valence electrons.

Section 8 Forming Chemical Bonds

Guillaume Favre



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