

| 6 Strong Acids                        | 6 Strong Bases                        |
|---------------------------------------|---------------------------------------|
| $\text{HClO}_4$ perchloric acid       | $\text{LiOH}$ lithium hydroxide       |
| $\text{HCl}$ hydrochloric acid        | $\text{NaOH}$ sodium hydroxide        |
| $\text{HBr}$ hydrobromic acid         | $\text{KOH}$ potassium hydroxide      |
| $\text{HI}$ hydroiodic acid           | $\text{Ca(OH)}_2$ calcium hydroxide   |
| $\text{HNO}_3$ nitric acid            | $\text{Sr(OH)}_2$ strontium hydroxide |
| $\text{H}_2\text{SO}_4$ sulfuric acid | $\text{Ba(OH)}_2$ barium hydroxide    |

## Section 8 3 Strength Of Acids And Bases

**ML Morrison**



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