

1. In the list below, circle the properties of acids and box the properties of bases.

Sour taste	Feel slippery	Contains hydroxide ions
Turns litmus blue	Reacts with metals to form H_2	Turns litmus red
Bitter taste	Contains hydrogen ions	Corrosive

2. Complete the chart below:

Acid-Base Definitions		
Type	Acid Definition	Base Definition
Arrhenius	Donates an H^+ ion in solution	Donates an OH^- ion in solution
Bronsted-Lowry	Donates an H^+ ion	Accepts an H^+ ion
Lewis	Accepts an electron pair	Donates an electron pair

3. What term(s) describe a substance that can act as an acid or a base? **Amphoteric or amphiprotic**
4. Neutralization reactions always produce a **SALT** and **WATER**
5. If the $[H^+]$ is greater than the $[OH^-]$, the solution is acidic basic neutral
6. If the $[OH^-]$ is greater than the $[H^+]$, the solution is acidic basic neutral
7. If the $[H^+]$ is equal to the $[OH^-]$, the solution is acidic basic neutral
8. If the pH is 7, the solution is acidic basic neutral
9. If the pH is greater than 7, the solution is acidic basic neutral
10. If the pH is less than 7, the solution is acidic basic neutral
11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?



$$(19.2 \text{ g } HClO_3) \times (1 \text{ mol } HClO_3/84 \text{ g } HClO_3) \times (1 \text{ mol } LiOH/1 \text{ mol } HClO_3) = \underline{0.228 \text{ mol } LiOH}$$

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?



$$(132.4 \text{ g } HNO_3) \times (1 \text{ mol } HNO_3/63 \text{ g } HNO_3) \times (1 \text{ mol } Ca(OH)_2/2 \text{ mol } HNO_3) \times (74 \text{ g } Ca(OH)_2/1 \text{ mol } Ca(OH)_2) = \underline{77.76 \text{ g } Ca(OH)_2}$$

Section 8 3 Properties Of Acids And Base Answers

Kaplan Test Prep



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Workbook for Organic Chemistry Jerry Jenkins, 2009-12-25 With authors who are both accomplished researchers and educators Vollhardt and Schore's Organic Chemistry is proven effective for making contemporary organic chemistry accessible introducing cutting edge research in a fresh student friendly way A wealth of unique study tools help students organize and understand the substantial information presented in this course And in the sixth edition the themes of understanding reactivity mechanisms and synthetic analysis to apply chemical concepts to realistic situations has been strengthened New applications of organic chemistry in the life sciences industrial practices green chemistry and environmental monitoring and clean up are incorporated This edition includes more than 100 new or substantially revised problems including new problems on synthesis and green chemistry and new challenging problems **Basic Concepts of Chemistry** Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance Physical Science Anted, Grades 6-9 David W. Appenbrink, 1986 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Understanding Chemistry* Fred M. Dewey, 1994 **Disha 30 Chapter-wise, Topic-wise & Skill-wise CBSE Class 10 Science Previous Year Solved Papers (2013 - 2025) & Sample Papers |PYQ, Notes | CBSE Question Bank Class 10 | 3rd Edition**, The 3rd updated edition of the book Disha 30 Chapter wise Topic wise Skill wise CBSE Question Bank Class 10 Science Previous Year Solved Papers 2013 2025 includes Solved papers of past 13 years along with CBSE Sample Papers Salient Features Solved papers of past 13 years along with CBSE Sample Papers For the first time ever a 3 Level division of the Solved Questions is presented in a Book Chapter wise Topic wise and Skill wise The Skill wise division divides the questions into Knowledge Understanding Application Analysis The Book is divided into 13 Chapters which are further divided into 52 Topics as per the NCERT Book covering 960 Questions All Questions pertaining to a Topic are provided here The Book includes 26

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