

Chapter 6 Chemical Bonds

Section 6.2 Covalent Bonding**(pages 166–169)**

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

Reading Strategy **page 166**

Relating Text and Visuals As you read the section, look closely at Figure 5. Complete the table for describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the Skills and Reference Handbook at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

Covalent Bonds **(pages 166–169)**

- Describe a covalent bond. _____

- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
 - electron dot
 - structural formula
 - space-filling
 - electron cloud
- Describe a molecule. _____

- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. _____
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
 - zero
 - one
 - two
 - three

Section 6 2 Covalent Bonding Answer Key

Tobias Bleicker

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Section 6 2 Covalent Bonding Answer Key:

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Chemistry David W. Oxtoby, Norman H. Nachtrieb, Wade A. Freeman, 1994 **Problems in Chemistry, Second Edition** Daley, 1988-02-19 **Descriptive Inorganic Chemistry** Geoff Rayner-Canham, Geoffrey Rayner-Canham, Tina Overton, 2009-12-28

This bestselling text introduces descriptive inorganic chemistry in a less rigorous, less mathematical way. The book uses the periodic table as a basis for understanding chemical properties and uncovering relationships between elements in different groups. Rayner-Canham and Overton's text also familiarizes students with the historical background of inorganic chemistry as well as with its crucial applications, especially in regard to industrial processes and environmental issues, resulting in a comprehensive appreciation and understanding of the field and the role it will play in their fields of further study.

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02

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A-Level Chemistry E. N. Ramsden, 2000

This textbook has been updated to cover the new specifications for AS and A2 Chemistry and improved with new features and rewritten material to enhance learning and increase accessibility. It covers all the main specifications for the English and Welsh Awarding Bodies and should be particularly suitable for students approaching A Level from GCSE Science Double Award. This answer key is designed to support the core book and contains suggested answers, worked solutions to the checkpoints and examination questions in the core book, also synoptic questions for further practice, complete with suggested answers and worked solutions to help develop confidence.

Objective Chemistry For IIT Entrance Alok Mittal, 2002

The book enables students to thoroughly master pre-college chemistry and helps them to prepare for various entrance screening tests with skill and confidence. The book thoroughly explains the following physical chemistry with detailed concepts and numerical problems; organic chemistry with more chemical equations and

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Quantitative Chemical Analysis Daniel C. Harris, 2010-04-30 QCA is the bestselling textbook of choice for analytical chemistry It offers a modern portrait of the techniques of chemical analysis backed by a wealth of real world applications This edition features new coverage of spectroscopy and statistics new pedagogy and enhanced lecturer support

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Photoinduced Molecular Dynamics in Solution Gianluca Levi, 2019-09-03 This book explores novel computational strategies for simulating excess energy dissipation alongside transient structural changes in photoexcited molecules and accompanying solvent rearrangements It also demonstrates in detail the synergy between theoretical modelling and ultrafast experiments in unravelling various aspects of the reaction dynamics of solvated photocatalytic metal complexes Transition metal complexes play an important role as photocatalysts in solar energy conversion and the rational design of metal based photocatalytic systems with improved efficiency hinges on the fundamental understanding of the mechanisms behind light induced chemical reactions in solution Theory and atomistic modelling hold the key to uncovering these ultrafast processes Linking atomistic simulations and modern X ray scattering experiments with femtosecond time resolution the book highlights previously unexplored dynamical changes in molecules and discusses the development of theoretical and computational frameworks capable of interpreting the underlying ultrafast phenomena

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