Section 5.2 The Modern Periodic Table (pages 130–138)

This section explains the organization of the modern periodic table and discusses the general properties of nectals, nonnectals, and nectaliside.

Reading Strategy (page 199)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132–135. As you read, write answers to your questions. For more information on this reading studgey, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Owestions About the Periodic Tuble	
Disestion	Answer
Disclants might ask: Hitself deed stome: Many mass?	Allestic chase to a veine that disponds on the distribution at an element's lookuped in nature and the impasse of those trubges.
Miley are loss portion of elements policied before the male hody of the false?	Their service of interments are placed factors the fallor to make the latter many inequals.

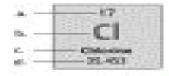
Efter possible greations Why are there has immitating systems to the optional FWhy is parted 7 incomplate?

The Periodic Law (pages 101-100)

- Is the following sentence true or false? In the modern periodic table, elements are arranged by increasing number of protons.
- Explain why the number of elements per period varies. The number of elements per period varies because the number of available orbitals increases from energy level to energy level.
- Proportion of observerts repeat in a productable way releast atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the

Atomic Mass tone 1040

 Label the four types of information supplied for chlorine in the diagram.



of Alcorety marehor

Element symbol

Element ourse

A Market Press

Section 5 2 The Modern Periodic Table Key

JS Bruner

Section 5 2 The Modern Periodic Table Key:
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Section 5 2 The Modern Periodic Table Key Introduction

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