



FISSION AND FUSION OF ATOMIC NUCLEI

Section Review

Objectives

1. Describe fission and fusion reactions.
2. Explain the conditions for the synthesis of superheavy elements.
3. Compare fusion reactions and fission reactions.

Vocabulary

- | | |
|---------------------------|----------------------------|
| 1. fusion | 12. fusion reaction |
| 2. fusion reaction | 13. fusion |

Part A Completion

Write the equation below in effect, substituting the correct numbers and symbols for the missing values. Each blank must represent a different particle involved.

1. Uranium-235, $^{235}_{92}\text{U}$, undergoes fission to produce barium-141, $^{141}_{56}\text{Ba}$, krypton-92, $^{92}_{36}\text{Kr}$, and three neutrons. The equation is:

$$^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{141}_{56}\text{Ba} + ^{92}_{36}\text{Kr} + 3^1_0\text{n}$$
2. Uranium-235, $^{235}_{92}\text{U}$, undergoes fission to produce barium-141, $^{141}_{56}\text{Ba}$, krypton-92, $^{92}_{36}\text{Kr}$, and three neutrons. The equation is:

$$^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{141}_{56}\text{Ba} + ^{92}_{36}\text{Kr} + 3^1_0\text{n}$$
3. Uranium-235, $^{235}_{92}\text{U}$, undergoes fission to produce barium-141, $^{141}_{56}\text{Ba}$, krypton-92, $^{92}_{36}\text{Kr}$, and three neutrons. The equation is:

$$^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{141}_{56}\text{Ba} + ^{92}_{36}\text{Kr} + 3^1_0\text{n}$$
4. Uranium-235, $^{235}_{92}\text{U}$, undergoes fission to produce barium-141, $^{141}_{56}\text{Ba}$, krypton-92, $^{92}_{36}\text{Kr}$, and three neutrons. The equation is:

$$^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{141}_{56}\text{Ba} + ^{92}_{36}\text{Kr} + 3^1_0\text{n}$$
5. Uranium-235, $^{235}_{92}\text{U}$, undergoes fission to produce barium-141, $^{141}_{56}\text{Ba}$, krypton-92, $^{92}_{36}\text{Kr}$, and three neutrons. The equation is:

$$^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{141}_{56}\text{Ba} + ^{92}_{36}\text{Kr} + 3^1_0\text{n}$$

14. **fusion**
15. **fusion reaction**
16. **fusion**
17. **fusion**
18. **fusion reaction**
19. **fusion**
20. **fusion**
21. **fusion**
22. **fusion**
23. **fusion**
24. **fusion**
25. **fusion**

Part B True-False

Write **TRUE** or **FALSE** next to each statement. If the statement is false, write why it is false.

1. **TRUE** Fusion reactions are exothermic.
2. **FALSE** Fusion reactions are exothermic.
3. **FALSE** Fusion reactions are exothermic.
4. **FALSE** Fusion reactions are exothermic.
5. **FALSE** Fusion reactions are exothermic.
6. **FALSE** Fusion reactions are exothermic.
7. **FALSE** Fusion reactions are exothermic.
8. **FALSE** Fusion reactions are exothermic.
9. **FALSE** Fusion reactions are exothermic.
10. **FALSE** Fusion reactions are exothermic.
11. **FALSE** Fusion reactions are exothermic.
12. **FALSE** Fusion reactions are exothermic.
13. **FALSE** Fusion reactions are exothermic.
14. **FALSE** Fusion reactions are exothermic.
15. **FALSE** Fusion reactions are exothermic.
16. **FALSE** Fusion reactions are exothermic.
17. **FALSE** Fusion reactions are exothermic.
18. **FALSE** Fusion reactions are exothermic.
19. **FALSE** Fusion reactions are exothermic.
20. **FALSE** Fusion reactions are exothermic.
21. **FALSE** Fusion reactions are exothermic.
22. **FALSE** Fusion reactions are exothermic.
23. **FALSE** Fusion reactions are exothermic.
24. **FALSE** Fusion reactions are exothermic.
25. **FALSE** Fusion reactions are exothermic.

Section 4 3 Niomes Section Review Answers

Lei Huang



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Table of Contents Section 4 3 Niomes Section Review Answers

1. Understanding the eBook Section 4 3 Niomes Section Review Answers
 - The Rise of Digital Reading Section 4 3 Niomes Section Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 4 3 Niomes Section Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 4 3 Niomes Section Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 4 3 Niomes Section Review Answers
 - Personalized Recommendations
 - Section 4 3 Niomes Section Review Answers User Reviews and Ratings
 - Section 4 3 Niomes Section Review Answers and Bestseller Lists
5. Accessing Section 4 3 Niomes Section Review Answers Free and Paid eBooks
 - Section 4 3 Niomes Section Review Answers Public Domain eBooks
 - Section 4 3 Niomes Section Review Answers eBook Subscription Services
 - Section 4 3 Niomes Section Review Answers Budget-Friendly Options

6. Navigating Section 4 3 Niomes Section Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 4 3 Niomes Section Review Answers Compatibility with Devices
 - Section 4 3 Niomes Section Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 4 3 Niomes Section Review Answers
 - Highlighting and Note-Taking Section 4 3 Niomes Section Review Answers
 - Interactive Elements Section 4 3 Niomes Section Review Answers
8. Staying Engaged with Section 4 3 Niomes Section Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 4 3 Niomes Section Review Answers
9. Balancing eBooks and Physical Books Section 4 3 Niomes Section Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 4 3 Niomes Section Review Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Section 4 3 Niomes Section Review Answers
 - Setting Reading Goals Section 4 3 Niomes Section Review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 4 3 Niomes Section Review Answers
 - Fact-Checking eBook Content of Section 4 3 Niomes Section Review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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