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CLASSIFYING CHEMICAL REACTIONS

Classify the following chemical reactions as one of the following: synthesis, decomposition, single-replacement, double-replacement, or combustion.

$2H_0 + O_0 \rightarrow 2H_0O$	Caco, Cao + Co,	
3K + 3H ₂ O → H ₄ + 3KOH	CHOO, • 2HCI CHOL • HLCO.	
290010 ₉ 28001 + 30 ₉	Ca + 2HCl CaCl ₂ + H ₂	
280 _e + 0 _e 280 _e	PERNOJ, - SHCI - PECL - SHEED,	
$Ov + 2AgNO_y \rightarrow 2Ag + OutyOy$	2Na • Cl _e 2NeCli	
2H ₂ O 2H ₂ - O ₂	CH ₆ + 20 ₅ → CO ₃ + 24 ₅ O	
$PbONO_{\mathcal{O}_{S}} = 290 \rightarrow PbI_{b} = 29000_{b}$	H,CO, CO, - H,O	
2Fe + O₂ → 2FeO	Zn + 2HCt H _b + ZnCt _b	
HCI + NaOH → NaOI + HOH	25% + 254,0 25%0H + H ₂	

Physical Science If8767 Classifying Chemical Reactions

R Barnett

Physical Science If8767 Classifying Chemical Reactions:

Decoding **Physical Science If8767 Classifying Chemical Reactions**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "Physical Science If8767 Classifying Chemical Reactions," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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