# WORD EQUATIONS Name Write and balance the following chemical equations. Hydrogen plus oxygen vield water. Nitrogen plus hydrogen yield ammonia. Aluminum bromide plus chlorine yield aluminum chloride and bromine. Hydrochloric acid plus sodium hydroxide yield sodium chloride plus water. Iron plus lead (II) sulfate react forming iron (II) sulfate plus lead. Potassium chlorate when heated produces potassium chloride plus oxygen gas. Sulfuric acid decomposes to form sulfur trioxide gas plus water. Sodium oxide combines with water to make sodium hydroxide.

- Potassium lodide reacts with bromine forming potassium bromide plus lodine.
- Sodium phosphate reacts with calcium nitrate to produce sodium nitrate plus 10. calcium phosphate.
- 11. Zinc reacts with iron (III) chloride yielding zinc chloride plus iron precipitate.
- 12. Ammonium carbonate and magnesium sulfate react to yield ammonium sulfate plus magnesium carbonate.
- Phosphoric acid plus calcium hydroxide react forming solid calcium phosphate plus 13. water.
- 14. Aluminum plus oxygen gas form aluminum oxide under certain conditions.
- 15. Nitrogen gas plus oxygen gas react and form dinitrogen pentoxide.

# **Physical Science If8767 Word Equations Answers**

**Julia Schneider** 

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Chemistry Equations & Answers Speedy Publishing,2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to students even the best ones However a chemistry equations and answers study guide can help the student When used as a quick reference guide it can be used often to determine the formulas needed for various questions. The astute student can cleverly devise ways to make the guide useful for test questions or other circumstances requiring one of the many chemistry equations. Equations of Mathematical Physics Andrej Nikolaevi? Tikhonov (Mathematician), Aleksandr Andreevich Samarski?, 1990-01-01 Mathematical physics plays an important role in the study of many physical processes hydrodynamics elasticity and electrodynamics to name just a few Because of the enormous range and variety of problems dealt with by mathematical physics this thorough advanced undergraduate or graduate level text considers only those problems leading to partial differential equations. The authors two well known Russian mathematicians have focused on typical physical processes and the principal types of equations deailing with them Special attention is paid throughout to mathematical formulation rigorous solutions and physical interpretation of

the results obtained Carefully chosen problems designed to promote technical skills are contained in each chapter along with extremely useful appendices that supply applications of solution methods described in the main text At the end of the book a helpful supplement discusses special functions including spherical and cylindrical functions The Ultimate Chemical Equations Handbook, Teacher Edition Jane D. Smith, George R. Hague, 2011 Basic equations of engineering science William Frank Hughes, E. W. Gaylord, 1975

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