

Koliko je kubnih cm 92% sumporne kiseline
(gustine=1.184g/cm³) potrebno za pripremanje 5dm³ 20%
rastvora H₂SO₄ čija je gustina 1.14 g/cm³
Hvala unapred!

Kreneš od rastvora koji imaš.

$$m = V \times \text{gustina}$$

$$m = 5000 \text{ cm}^3 \times 1.14 \text{ g/cm}^3 = 5700 \text{ g rastvora H}_2\text{SO}_4$$

$$100 \text{ g} : 20 \text{ g čiste H}_2\text{SO}_4 = 5700 \text{ g} : m \text{ čiste H}_2\text{SO}_4$$

$$m = 1140 \text{ g čiste H}_2\text{SO}_4$$

Potrebni rastvor:

$$100 \text{ g} : 92 \text{ g čiste H}_2\text{SO}_4 = m_{\text{rastvora H}_2\text{SO}_4} : 1140 \text{ g čiste H}_2\text{SO}_4$$

$$m = 1239,13 \text{ g rastvora H}_2\text{SO}_4$$

$$V = m/\text{gustina}$$

$$V = 1239,13 \text{ g} / 1,184 \text{ g/cm}^3$$

$$V = 1046,5 \text{ cm}^3 \text{ rastvora H}_2\text{SO}_4$$

Trebalo bi da ide ovako. Proveri rešenje

1. 1.Koliko atoma gvozdja se nalazi u 5,6g Fe
- 2.Izracunaj kolicinu, masu i zapreminu $1,5 \cdot 10^{23}$ molekula fluora! (to "10²³" znaci isto kao i 2 na kvadrat samo je iznad 23) 😊
- 3.Kolika je masa CO koji se nalazi u balonu zapremine 5,6 dm na 3(kubni)
- 4.Koju zapreminu zauzima 32g sumpor-(IV) oksida
- 5.U balonu se nalaze 0,2 mola kiseonika, $3 \cdot 10^{23}$ molekula azota i 1,12 dm na 3 (kubni) CO₂.Kolika je masa balona

To je prvih pet a inace ih ima 20



Reseni Zadaci Iz Hemije Za Srednju Skolu

M Carnoy



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