

# REPORT FOR EXPERIMENT 12

INSTRUCTOR

## Single Displacement Reactions

| Evidence of Reaction                                                          | Equations (to be completed)                                                           |
|-------------------------------------------------------------------------------|---------------------------------------------------------------------------------------|
| Describe any evidence of reaction; if no reaction was observed, write "None". | Write "No reaction", if no reaction was observed.                                     |
| 1. <u>spongy, dark red/black</u>                                              | $\text{Cu} + \text{AgNO}_3(\text{aq}) \rightarrow \text{Cu(NO}_3)_2 + 2\text{Ag}$     |
| 2. <u>dark</u>                                                                | $\text{Pb} + \text{Cu(NO}_3)_2(\text{aq}) \rightarrow \text{Pb(NO}_3)_2 + \text{Cu}$  |
| 3. <u>spongy, dark</u>                                                        | $\text{Zn} + \text{Pb(NO}_3)_2(\text{aq}) \rightarrow \text{Zn(NO}_3)_2 + \text{Pb}$  |
| 4. <u>none</u>                                                                | $\text{Zn} + \text{MgSO}_4(\text{aq}) \rightarrow \text{no reaction}$                 |
| 5. <u>none</u>                                                                | $\text{Cu} + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow \text{no reaction}$         |
| 6. <u>same color spongy being formed</u>                                      | $\text{Zn} + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow \text{ZnSO}_4 + \text{H}_2$ |

## QUESTIONS AND PROBLEMS

- Complete the following table by writing the symbols of the two elements whose reactivities are being compared in each test:

|                  | Tube Number |           |           |           |           |           |
|------------------|-------------|-----------|-----------|-----------|-----------|-----------|
|                  | 1           | 2         | 3         | 4         | 5         | 6         |
| Greater activity | <u>Cu</u>   | <u>Pb</u> | <u>Zn</u> | <u>Mg</u> | <u>H</u>  | <u>Zn</u> |
| Lesser activity  | <u>Ag</u>   | <u>Cu</u> | <u>Pb</u> | <u>Zn</u> | <u>Cu</u> | <u>H</u>  |

- Arrange Pb, Mg, and Zn in order of their activities, listing the most active first.

- (1) Mg
- (2) Zn
- (3) Pb

# Report For Experiment 12 Single Displacement Reactions

**RS Peters**



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