

5 - Periodicity Activity

1 2 3 4 5 6 7 8 9 10 11 12

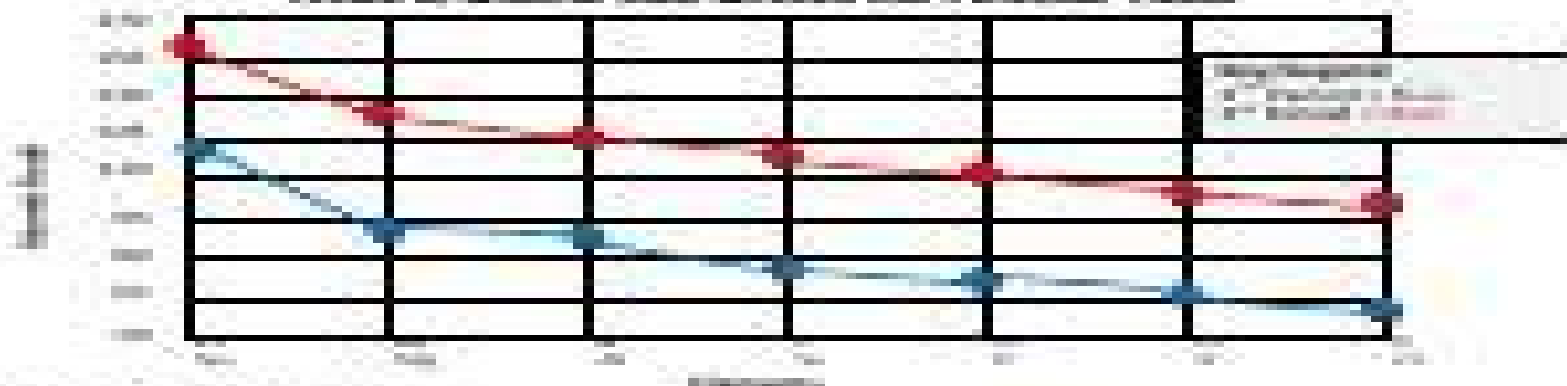
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the values of atomic property below:
2. Is a slight downward trend in this graph Period? If not, is the value graph:
3. Slight upward or not include irregularity
4. Downward irregularity.

| | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 |
|---|------|------|------|------|------|------|------|------|
| 1 | 1H | 2He | | | | | | |
| 2 | 3Li | 4Be | 5B | 6C | 7N | 8O | 9F | 10Ne |
| 3 | 11Na | 12Mg | 13Al | 14Si | 15P | 16S | 17Cl | 18Ar |
| 4 | 19K | 20Ca | 21Sc | 22Ti | 23V | 24Cr | 25Mn | 26Fe |
| 5 | 27Co | 28Ni | 29Cu | 30Zn | 31Ga | 32Ge | 33As | 34Se |
| 6 | 35Br | 36Kr | 37Rb | 38Sr | 39Y | 40Zr | 41Nb | 42Mo |
| 7 | 43Tc | 44Ru | 45Rh | 46Pd | 47Ag | 48Cd | 49In | 50Sn |

Now fill in the boxes below (1-12) below the graph:

Trend in Atomic Size across the Periodic Table



Shortened Trend Questions

1. Following the trend in atomic size across the periodic table: decrease
2. This across two rows there are more downward periodicity differences. They are all downward spelling differences, almost no irregularity, however.

Full trend Trend Questions

1. Shortened trend of the 1st ionization energy (IE) is a downward trend. The full trend in the periodic table is more or less the same as the periodic table: decrease
2. This across two rows there are more downward periodicity differences (a lot of differences) spelling the downward differences.

Periodicity Activity Okstate Answer Key

Daniel F McAuley



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