

Chapter 5 The Periodic Table

Section 5.2 The Modern Periodic Table

(pages 130–136)

This section explains the organization of the modern periodic table and discusses the general properties of metals, nonmetals, and metalloids.

Reading Strategy (page 130)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132–133. As you read, write answers to your questions. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Questions About the Periodic Table	
Question	Answer
Students might ask: What does atomic mass mean?	Atomic mass is a value that depends on the distribution of an element's isotopes in nature and the masses of those isotopes.
Why are two series of elements placed below the main body of the table?	Two series of elements are placed below the table to make the table more compact.

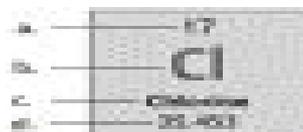
Other possible questions: Why are there two numbering systems for the columns? Why is period 7 incomplete?

The Periodic Law (pages 131–133)

1. Is the following sentence true or false? In the modern periodic table, elements are arranged by increasing number of protons.
_____ **true** _____
2. Explain why the number of elements per period varies. _____
_____ **The number of elements per period varies because the number of available orbitals increases from energy level to energy level.** _____
3. Properties of elements repeat in a predictable way when atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the _____.
_____ **periodic law** _____

Atomic Mass (page 134)

4. Label the four types of information supplied for chlorine in the diagram.



- a. _____ **Atomic number** _____ b. _____ **Element symbol** _____
 c. _____ **Element name** _____ d. _____ **Atomic mass** _____

Periodic Table Section 5 2 Answer

ML Morrison



Periodic Table Section 5 2 Answer:

Decoding **Periodic Table Section 5 2 Answer**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Periodic Table Section 5 2 Answer**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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