
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, please check that her responses match statements or facts covered in her presentation.

1. In developing his periodic table, Mendeleev's observations were
 - a. accurate enough.
 - b. accurate and systematic.
 - c. lacking.
 - d. **incomplete.**
2. Mendeleev's periodic table did not fit because he used only the
 - a. physical characteristics of elements for grouping together elements with similar
 - b. **physical characteristics.**
 - c. physical characteristics.
 - d. **chemical characteristics.**
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. holes.
 - b. unknown elements.
 - c. unstable elements.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on the
 - a. atomic number.
 - b. **atomic weight.**
 - c. atomic mass.
 - d. **atomic number.**
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. **compounds.**
 - c. **atoms.**
 - d. **elements.**

Periodic Law Section Quiz Answers

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