

CHAPTER 5: PERIODIC

The Periodic Law

SECOND PART

Student's Assistant Answered the following questions for the second part of the lab.

1. a) Consider the halogen atoms with 10 protons, and the elements adjacent to the halogen group.

- _____ a. The first is the atomic number?
- _____ b. The last is the atomic number?
- _____ c. Is the element in period 10 or another period, rather than period 10? Is the element number 10 the atomic number?

2. Consider the elements with electron configurations $1s^2 2s^2 2p^2$, $1s^2 2s^2 2p^3$.

- _____ a. The first element is the atomic number?
- _____ b. The last is the atomic number?
- _____ c. The group number is equal to $10 + 10 = 20$. Is it?

- _____ d. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ e. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.

- _____ f. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.

- _____ g. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ h. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.

3. Write the electron configuration for the first 10 elements of the periodic table. Is the element 10 the first element?

- _____ a. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ b. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ c. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ d. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ e. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ f. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ g. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ h. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ i. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.
- _____ j. The first element is the atomic number, $1, 2, 3, 4, 5, 6$.

Periodic Law Mixed Review Answer Key

SB Merriam



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