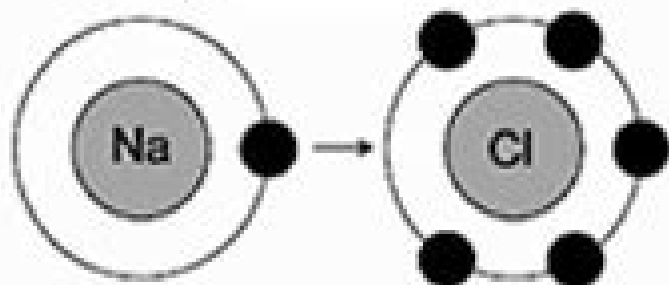
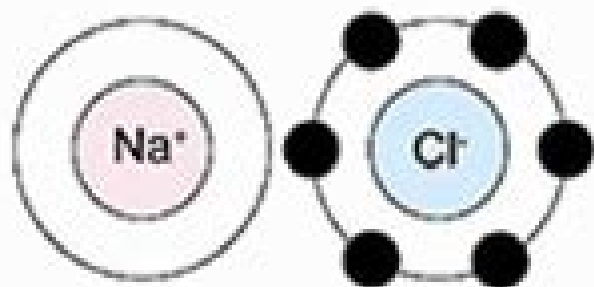


Covalent vs. Ionic Bonds

Ionic bond

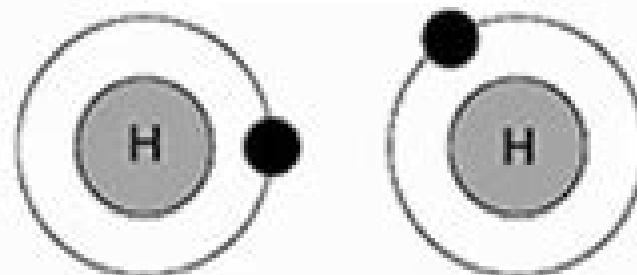


(a) The electron moves from Na to Cl

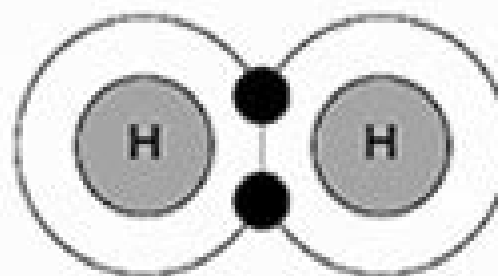


(b) Positively charged Na and negatively charged Cl attract each other and are electronically bonded together

Covalent bond



(c) Electrons cannot move independently



(d) The atoms form bonds by sharing electrons.

Reinforcement Ionic And Covalent Bonds Answers

**Kenneth Raymond Miller, Prentice Hall
(School Division)**



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