

Davis P2

# Periodic Table of the Elements

Handwritten notes:

- Valence Electrons
- groups (columns)
- periods (rows)
- Mass # = protons + neutrons
- metals
- non-metals
- metalloids

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1	H	He																
2	Li	Be										B	C	N	O	F	Ne	
3	Na	Mg										Al	Si	P	S	Cl	Ar	
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

<ul style="list-style-type: none"> <li>1: alkali metals</li> <li>2: alkaline earth metals</li> <li>3: transition metals</li> <li>4: noble gases</li> </ul>	<ul style="list-style-type: none"> <li>5: metalloids</li> <li>6: lanthanides</li> <li>7: actinides</li> </ul>	<ul style="list-style-type: none"> <li>8: halogens</li> <li>9: noble gases</li> <li>10: non-metals</li> </ul>	<ul style="list-style-type: none"> <li>11: solid</li> <li>12: liquid</li> <li>13: gas</li> </ul>
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# Reinforcement And Study Guide The Periodic Table

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