Figure 1 show a part of the periodic table.

7	2								13	14	1.5	16	1.7	7.8
								1		P		Q		+
	R	 											S	T
-		 -	-	-			_			-		-	-	-
		_		1		-		1			-	-	_	

Figure 1

Based on the Figure 1, answer the following questions.

- (a) What is the element represented by P?
 -(1 marks)
- (b) On the figure above, mark X in the box for an alkali metal (1 marks)
- (c) State one characteristic of element U. (1 marks)
- (d) Write the electron arrangement for S.
-(1 marks)
- (e) (i) Which element is not chemically reactive? (1 marks)
 - (ii)State one application of the element in (e)(i) (1 marks)
- (f) (i) Write the chemical equation for the reaction between R and Q
 - (1 marks)
 - (ii)State briefly the transfer of electron which occur during the reaction between R and Q.

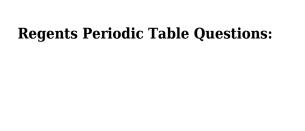
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3400400400400400400400400400400400400400	100	ALL PROPERTY OF THE PARTY OF

JAWAPAN

l (a)	Carbon	1
(b)	(any box in column for group 1)	1
(c)	Form coloured compounds	1
(d)	2.8.7	1
(e) (i) (ii)	T Fill electric bulb	1
(f) (i) (ii)	2R + Q ₁ → 2RQ R release two electrons to Q Q receive two electron from R	1 1 1
	***************************************	9

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