

2. Figure 1 show a part of the periodic table.

	1	2												13	14	15	16	17	18
1																			
2															P		Q		
3		R																S	T
4																			
5																			
6																			

Figure 1

Based on the Figure 1, answer the following questions.

- (a) What is the element represented by P? (1 marks)
- (b) On the figure above, mark X in the box for an alkali metal (1 marks)
- (c) State one characteristic of element U. (1 marks)
- (d) Write the electron arrangement for S. (1 marks)
- (e) (i) Which element is not chemically reactive? (1 marks)
- (ii) State one application of the element in (e)(i) (1 marks)
- (f) (i) Write the chemical equation for the reaction between R and Q (1 marks)
- (ii) State briefly the transfer of electron which occur during the reaction between R and Q. (2 marks)

JAWAPAN

1 (a)	Carbon	1
(b)	(any box in column for group 1)	1
(c)	Form coloured compounds	1
(d)	2.8.7	1
(e) (i)	T	1
(ii)	Fill electric bulb	1
(f) (i)	$2R + Q_2 \rightarrow 2RQ$	1
(ii)	- R release two electrons to Q - Q receive two electron from R	1 1
		9

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