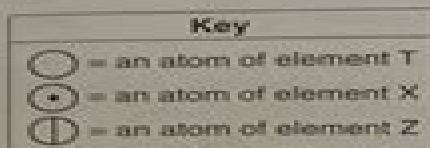


39 Given four particle models:



I



II



III



IV

Which two models can be classified as elements?

- (1) I and II  
(2) I and IV  
(3) II and III  
(4) II and IV

40 After being thoroughly stirred at  $10^{\circ}\text{C}$ , which mixture is heterogeneous?

- (1) 25.0 g of KCl and 100. g of  $\text{H}_2\text{O}$  unsat.  
(2) 25.0 g of  $\text{KNO}_3$  and 100. g of  $\text{H}_2\text{O}$  supersat.  
(3) 25.0 g of NaCl and 100. g of  $\text{H}_2\text{O}$  unsat.  
(4) 25.0 g of  $\text{NaNO}_3$  and 100. g of  $\text{H}_2\text{O}$  unsat.

41 Which two compounds are electrolytes?

- (1) KOH and  $\text{CH}_3\text{COOH}$   
(2) KOH and  $\text{C}_3\text{H}_{12}$   
(3)  $\text{CH}_3\text{OH}$  and  $\text{CH}_3\text{COOH}$   
(4)  $\text{CH}_3\text{OH}$  and  $\text{C}_3\text{H}_{12}$

42 Which statement explains why a  $\text{CO}_2$  molecule is nonpolar?

- (1) Carbon and oxygen are both nonmetals.  
(2) Carbon and oxygen have different electronegativities.  
(3) The molecule has a symmetrical distribution of charge.  
(4) The molecule has an asymmetrical distribution of charge.

43 Which temperature change indicates an increase in the average kinetic energy of the molecules in a sample?

- (1)  $15^{\circ}\text{C}$  to  $298\text{ K}$   $\uparrow$   
(2)  $37^{\circ}\text{C}$  to  $273\text{ K}$   $\downarrow$   
(3)  $305\text{ K}$  to  $0^{\circ}\text{C}$   $\downarrow$   
(4)  $355\text{ K}$  to  $25^{\circ}\text{C}$   $\downarrow$

44 Given the particle diagram:



Which substance at STP can be represented by this particle diagram? Solid

- (1)  $\text{N}_2$  (g)  
(2)  $\text{H}_2$  (g)  
(3) Mg (s)  
(4) Kr (g)

45 Which type of equilibrium exists in a sealed flask containing  $\text{Br}_2(\ell)$  and  $\text{Br}_2(\text{g})$  at  $298\text{ K}$  and  $1.0\text{ atm}$ ?

- (1) static phase equilibrium  
(2) static solution equilibrium  
(3) dynamic phase equilibrium  
(4) dynamic solution equilibrium

46 What are the products when potassium hydroxide reacts with hydrochloric acid?

- (1)  $\text{KH}(\text{s})$ ,  $\text{Cl}^+(\text{aq})$ , and  $\text{OH}^-(\text{aq})$   
(2)  $\text{K}(\text{s})$ ,  $\text{Cl}_2(\text{g})$ , and  $\text{H}_2\text{O}(\ell)$   
(3)  $\text{KCl}(\text{aq})$  and  $\text{H}_2\text{O}(\ell)$   
(4)  $\text{KOH}(\text{aq})$  and  $\text{Cl}_2(\text{g})$

47 In a titration, 20.0 milliliters of a  $0.150\text{ M}$   $\text{NaOH}(\text{aq})$  solution exactly neutralizes 24.0 milliliters of an  $\text{HCl}(\text{aq})$  solution. What is the concentration of the  $\text{HCl}(\text{aq})$  solution?

- (1)  $0.125\text{ M}$   
(2)  $0.180\text{ M}$   
(3)  $0.250\text{ M}$   
(4)  $0.360\text{ M}$

48 What fraction of a  $\text{Sr-90}$  sample remains unchanged after 87.3 years?

- (1)  $\frac{1}{2}$   
(2)  $\frac{1}{3}$   
(3)  $\frac{1}{4}$   
(4)  $\frac{1}{8}$

Handwritten calculation:  

$$3 = \frac{87.3}{29.14}$$

$$\frac{1}{2} = \frac{1}{8}$$

# Regents Chemistry Topic 4 Review Answers

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web napa white lithium grease manufacturer crc industries product code 1008003 095037 revision date 2017 october 09  
language english

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web category 1 osha defined hazards label elements 3 composition information on ingredients mixtures designates that a specific chemical identity and or percentage of composition has been withheld as a trade secret name napa 1066 white lithium grease us product l23 napa version 02

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web product identifier napa white lithium grease other means of identification product code no 79520 item 1006467  
recommended use lubricating grease recommended restrictions none known manufacturer importer supplier distributor  
information manufactured or sold by company name crc canada co address 2 1246 lorimar

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