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**Table A**  
**Standard Temperature and Pressure**

Name	Value	Unit
Standard Pressure	101.3 kPa 1 atm	kilopascals atmospheres
Standard Temperature	273 K 0 °C	kelvin degrees Celsius

- 7) Which sample of matter has the greatest increase between its boiling and STP?

- (1)  $\text{H}_2\text{O}$  (2)  $\text{C}_2\text{H}_6$  (3)  $\text{C}_2\text{H}_4$  (4)  $\text{C}_2\text{H}_2$   
(5)  $\text{H}_2\text{SO}_4$  (6)  $\text{C}_2\text{H}_6$  (7)  $\text{C}_2\text{H}_4$  (8)  $\text{C}_2\text{H}_2$

The melting points and boiling points of five substances at standard pressure are listed in the table below.

**Melting Points and Boiling Points of Five Substances**

Substance	Melting Point (°C)	Boiling Point (°C)
$\text{HCl}$	-109	-85
$\text{H}_2\text{O}$	0	100
$\text{F}_2$	-120	-85
$\text{Br}_2$	-7.3	58.8
$\text{I}_2$	113.7	184.3

- 8) Which substance in the table has a liquid range of 10 °C?



- 9) Convert the liquid range of the substance in the table to kelvin.

# Regents Chemistry New York Lab Manual

**M Tight**



## **Regents Chemistry New York Lab Manual:**

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