

## Electrochemistry Class 12 Important Questions Very Short Answer Type

Question 1.

What is meant by 'limiting molar conductivity'? (All India 2010)

Answer:

The molar conductivity of a solution at infinite dilution is called limiting molar conductivity and is represented by the symbol  $\Lambda_{\infty}$ .

Question 2.

Express the relation between conductivity and molar conductivity of a solution held in a cell. (Delhi 2011)

Answer:

$\Lambda_{\infty} = \kappa C$  = Conductivity Concentration

Question 3.

What is the effect of catalyst on:

- (i) Gibbs energy ( $\Delta G$ ) and
- (ii) activation energy of a reaction? (Delhi 2017)

Answer:

- (i) There will be no effect of catalyst on Gibbs energy.
- (ii) The catalyst provides an alternative pathway by decreasing the activation energy of a reaction.

Question 4.

What is the effect of adding a catalyst on

- (a) Activation energy ( $E_a$ ), and
- (b) Gibbs energy ( $\Delta G$ ) of a reaction? (All India 2017)

Answer:

- (a) On adding catalyst in a reaction, the activation energy reduces and rate of reaction is fastened.
- (b) A catalyst does not alter Gibbs energy ( $\Delta G$ ) of a reaction.

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Question 5.

Two half cell reactions of an electrochemical cell are given below :



Construct the redox equation from the two half cell reactions and predict if this reaction favours formation of reactants or product shown in the equation. (All India 2009)

Answer:

The reactions can be represented at anode and at cathode in the following ways :

# Redox And Electrochemistry Regents Short Answer Questions

**DJ Losen**



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