

**GRADE 10 PHYSICAL SCIENCES**

**NOVEMBER EXAMINATION PAPER 2: CHEMISTRY**

**SECTION A**

**QUESTION 1**

Various options are provided as possible answers to the following questions. Choose the correct answer and write **ONLY** the LETTER (A-D) next to the QUESTION NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D syngas

1.2 Element with the electron configuration:  $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B Na<sub>2</sub>S
- C NaS<sub>2</sub>
- D Na<sub>2</sub>S<sub>2</sub>

1.4 Strongest intermolecular force found in water?

- A Ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed
- B vibrate against each other
- C move freely and rarely touch
- D move over one another

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a:



- A noble gas
- B halogen
- C alkali earth-metal
- D alkali metal

# Physical Science Paper 2 Grade 10 November Exemplar

**RJ Alexander**



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## **Table of Contents Physical Science Paper 2 Grade 10 November Exemplar**

1. Understanding the eBook Physical Science Paper 2 Grade 10 November Exemplar
  - The Rise of Digital Reading Physical Science Paper 2 Grade 10 November Exemplar
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper 2 Grade 10 November Exemplar
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Paper 2 Grade 10 November Exemplar
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper 2 Grade 10 November Exemplar
  - Personalized Recommendations
  - Physical Science Paper 2 Grade 10 November Exemplar User Reviews and Ratings
  - Physical Science Paper 2 Grade 10 November Exemplar and Bestseller Lists
5. Accessing Physical Science Paper 2 Grade 10 November Exemplar Free and Paid eBooks

- Physical Science Paper 2 Grade 10 November Exemplar Public Domain eBooks
  - Physical Science Paper 2 Grade 10 November Exemplar eBook Subscription Services
  - Physical Science Paper 2 Grade 10 November Exemplar Budget-Friendly Options
6. Navigating Physical Science Paper 2 Grade 10 November Exemplar eBook Formats
- ePub, PDF, MOBI, and More
  - Physical Science Paper 2 Grade 10 November Exemplar Compatibility with Devices
  - Physical Science Paper 2 Grade 10 November Exemplar Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Paper 2 Grade 10 November Exemplar
  - Highlighting and Note-Taking Physical Science Paper 2 Grade 10 November Exemplar
  - Interactive Elements Physical Science Paper 2 Grade 10 November Exemplar
8. Staying Engaged with Physical Science Paper 2 Grade 10 November Exemplar
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Paper 2 Grade 10 November Exemplar
9. Balancing eBooks and Physical Books Physical Science Paper 2 Grade 10 November Exemplar
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Paper 2 Grade 10 November Exemplar
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Paper 2 Grade 10 November Exemplar
- Setting Reading Goals Physical Science Paper 2 Grade 10 November Exemplar
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Paper 2 Grade 10 November Exemplar
- Fact-Checking eBook Content of Physical Science Paper 2 Grade 10 November Exemplar
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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