



GRADE 10 PHYSICAL SCIENCES

NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

SECTION A

QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write ONLY the LETTER (A-D) next to the QUESTION/NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D syngas

1.2 Element with the electron configuration $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B Na_2S
- C NaS_2
- D Na_2S_2

1.4 Strongest intermolecular force found in water?

- A ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed
- B vibrate against each other
- C move freely and rarely touch
- D move over one another

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a

$1s^2$	2s
$2s^2 2p^4$	3s
$3s^2$	3p
$3d^5$	4s

- A noble gas
- B halogen
- C alkali earth metal
- D alkali metal

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Table of Contents Physical Science Grade Paper 2 2012 November

1. Understanding the eBook Physical Science Grade Paper 2 2012 November
 - The Rise of Digital Reading Physical Science Grade Paper 2 2012 November
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade Paper 2 2012 November
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade Paper 2 2012 November
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade Paper 2 2012 November

- Personalized Recommendations
 - Physical Science Grade Paper 2 2012 November User Reviews and Ratings
 - Physical Science Grade Paper 2 2012 November and Bestseller Lists
5. Accessing Physical Science Grade Paper 2 2012 November Free and Paid eBooks
 - Physical Science Grade Paper 2 2012 November Public Domain eBooks
 - Physical Science Grade Paper 2 2012 November eBook Subscription Services
 - Physical Science Grade Paper 2 2012 November Budget-Friendly Options
 6. Navigating Physical Science Grade Paper 2 2012 November eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Grade Paper 2 2012 November Compatibility with Devices
 - Physical Science Grade Paper 2 2012 November Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade Paper 2 2012 November
 - Highlighting and Note-Taking Physical Science Grade Paper 2 2012 November
 - Interactive Elements Physical Science Grade Paper 2 2012 November
 8. Staying Engaged with Physical Science Grade Paper 2 2012 November
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade Paper 2 2012 November
 9. Balancing eBooks and Physical Books Physical Science Grade Paper 2 2012 November
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade Paper 2 2012 November
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science Grade Paper 2 2012 November
 - Setting Reading Goals Physical Science Grade Paper 2 2012 November
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science Grade Paper 2 2012 November

- Fact-Checking eBook Content of Physical Science Grade Paper 2 2012 November
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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