# CHEMISTRY TEST

60 min . Para No calculator

Directions: Each set of lettered choices below refers to the numbered questions, formulas, or statements immediately following it. Select the one lettered choice that best answers each question or best fits each formula or statement and then blacken the corresponding space on the answer sheet. A choice may be used once, more than once, or not at all in each set.

#### Questions 1-4

- (A) Thermometer
- (B) Separatory funnel
- (C) Flame spectrophotometer
- (D) Buretse
- (E) Volumetric flask
- Used to study the characteristic emission lines produced by excited atoms of elements
- Used for the preparation of solutions of specific concentrations when a solid is used as the solute
- Commonly used for removing the water layer from a system containing oil and water
- 4. Used for volume measurements during titration

#### Questions 5-9

- CAN Linear.
- (B) Bent (V-shaped)
- (C) Trigonal, planar
- (D) Pyramidal
- (E) Regular tetrahedral

From the list above, select the shape that describes each of the following.

- 5. H,O
- 6. BI
- 7. NH.
- S. Ber.
- Q. NIH.

#### Questions 10-12

- (A) NaHCO<sub>2</sub>
- (B) H,50.
- (C) CH.
- (D) Ca(OH)<sub>2</sub>
- (E) C.H.
- 10. A significant constituent of baking powders
- 11. A constituent of gasoline
- 12. A major constituent of natural gas

#### Questions 13-15

- (A)  $3 O_2(g) = 2 O_2(g)$
- (B) OH' + H<sub>2</sub>O' = 2 H<sub>2</sub>O
- (C) BaCl<sub>2</sub>·2 H<sub>2</sub>O(s) △ BaCl<sub>2</sub>(s) + 2 H<sub>2</sub>O(g)
- (D) Ca2 + CO-2 - CaCO-(a)
- (E) Fe + Cu2 = Fe2 + Cu
- 13. Represents an oxidation-reduction reaction
- Involves the formation of an ionic precipitate from a solution
- 15. Represents a Brönsted acid-base reaction

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