

Groups and Periods on the Periodic Table Answers

- 1. Write the group numbers in the boxes along the top of the periodic table.
- 2. Write the period numbers in the boxes down the side of the periodic table.

	1	2											3	4	5	6	7	
1								H										He
2	Li	Be											В	c	N	o	F	Ne
3	Na	Mg											Al	Si	P	S	CI	Ar
4	K	Ca	S-c	Ťi	v	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sm	Sb	Te	Ĭ.	же
6	Cs	Ва	arra .	Hif	Ta	w	Re	Os	lir	Pt	Au	Hg	D.	Pb	Bi	Po	At	Rn
7	Fr	Ra		Rf	οъ	S-g	Bh	Hs	Mt	Ds	Rg	Cn	Nh	FI	Mic	Lv	Ts	Og



Groups and Periods on the Periodic Table Answers

Elements in the same group have similar properties.

Lithium (Li) and potassium (K) are metals. They are solids at room temperature, have low melting points compared to other metals and are very reactive.

3. Name two other elements that have similar properties to lithium and potassium.

Any two from:

- sodium (Na)
- · rubidium (Rb)
- caesium (Cs)
- francium (Fr)

Neon (Ne) and xenon (Xe) are non-metals. They are gases at room temperature and are unreactive.

4. Name two other elements that have similar properties to neon and xenon.

Any two from:

- · helium (He)
- · argon (Ar)
- krypton (Kr)
- · radon (Rn)

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