

4. Which type of radiation – alpha, beta, or gamma:

a. Results in the greatest change in atomic number? Why?

\_\_\_\_\_

b. Results in the least change in atomic number? Why?

\_\_\_\_\_

c. Produces the greatest change in mass number? Why?

\_\_\_\_\_

d. Produces the least change in mass number? Why?

\_\_\_\_\_

5. Complete the following nuclear reactions:



6. When isotopes lanthanide-213 emits an alpha particle:

a. Write out the nuclear equation:

b. Which is the parent element? \_\_\_\_\_

c. Which is the daughter element? \_\_\_\_\_

d. Write out the nuclear equation if the lanthanide, lanthanid, emits a beta particle:

# Regents Prep Nuclear Chemistry Topic 12 Answers

**L Cohen**



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