

# Bonding and Reactions

## Study Guide

### 10 pages including KEY

#### Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts on the exam. This packet highlights key concepts and their correct answers on the exam and should be used as a guide to the material you should know.

#### B. VOCABULARY

Atom	Electron	Electron transfer	Electron transfer
Atomic mass	Electron	Electron transfer	Electron transfer
Atomic number	Electron	Electron transfer	Electron transfer
Chemical bond	Electron	Electron transfer	Electron transfer
Chemical reaction	Electron	Electron transfer	Electron transfer

#### C. READING

1. Why do atoms bond together?
2. What is a covalent bond?
3. What kind of electrons form covalent bonds?
4. What is an ionic bond?
5. What kind of electrons form ionic bonds?
6. Why don't noble gases bond together with other elements?

Write the formula for the following compounds with multiple valence numbers.

1. Copper (II) chloride
2. Copper (II) chloride

Write the following compounds with multiple valence numbers.

1. Iron
2. Iron

#### REACTIONS

What happens in an oxidation-reduction reaction?

What does an oxidation-reduction reaction involve?

What is an oxidation-reduction reaction?

What is an oxidation-reduction reaction?

What is the law of conservation of mass?

What is the law of conservation of mass?

A coefficient is a number placed in front of a chemical formula to indicate the number of molecules of that substance that are involved in a chemical reaction. It is used to balance the equation.

What is a chemical reaction? A chemical reaction is a process in which one or more substances are converted into one or more new substances. The substances that enter the reaction are called reactants, and the substances that are produced are called products.

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# Physical Science Bonding And Chemical Reactions Test

**Didier Musso**



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