



MERCY JUNIOR COLLEGE  
Grade 11  
Physical Science  
Finals Exam

NAME: \_\_\_\_\_

YEAR LEVEL & SECTION: \_\_\_\_\_

TEACHER'S NAME: \_\_\_\_\_

PROCTOR'S NAME: \_\_\_\_\_

1. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol  $e^-$  or  $\beta^-$ , with a negative elementary electric charge.
  - Neutron
  - Electron
  - Proton
  - Atom
- It is a subatomic particle, symbol  $p$  or  $p^+$ , with a positive electric charge of  $+1e$  elementary charge and mass slightly less than that of a neutron.
  - Neutron
  - Electron
  - Proton
  - Atom
- It is a subatomic particle, symbol  $n$  or  $n^0$ , with no net electric charge and a mass slightly larger than that of a proton.
  - Neutron
  - Electron
  - Proton
  - Atom
- Is this an example of ionic bonding?



carbon dioxide

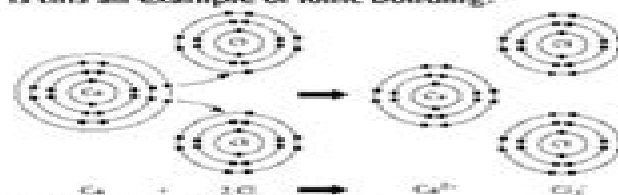
- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- Is this an example of ionic bonding?



lithium fluoride

- Yes, it is ionic bonding.
  - No, it is of another type of chemical bonding.
  - There is no chemical bonding present.
  - This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
    - Electron Configuration
    - Periodic Table of Elements
    - Orbital Models
    - Energy Levels
  - Is this an example of ionic bonding?



- Yes, it is ionic bonding.
  - No, it is of another type of chemical bonding.
  - There is no chemical bonding present.
  - This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
    - Octet Rule
    - Valence Electrons
    - Atomic Orbitals
    - Chemical Bonds
  - It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
    - Neutron
    - Electron
    - Proton
    - Atom
  - It is a type of chemical bond that involves the electrostatic attraction between oppositely

# Physical Science Grade 11 Final Examination 2014 Scope

**Judd E. Hollander**



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worldviews influence the kinds of mathematical instruments we construct and accept In addition contributors examine how concepts of rhetoric such as analogy and visibility have been employed in mathematical and scientific reasoning including in the theorems of mathematical physicists and the geometrical diagramming of natural scientists Challenging academic orthodoxy these scholars reject a math equals truth reduction in favor of a more constructivist theory of mathematics as dynamic evolving and powerfully persuasive By bringing these disparate lines of inquiry into conversation with one another Arguing with Numbers provides inspiration to students established scholars and anyone inside or outside rhetorical studies who might be interested in exploring the intersections between the two disciplines In addition to the editors the contributors to this volume are Catherine Chaput Crystal Broch Colombini Nathan Crick Michael Dreher Jeanne Fahnestock Andrew C Jones Joseph Little and Edward Schiappa

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