

QUESTION 12

- (a)
- The number of photoelectrons emitted per second is directly proportional to the intensity of radiation.
 - The energy of the emitted electrons depend on the frequency of the incident radiation.
 - There exists a threshold frequency for all surfaces, below which no electrons are emitted no matter the intensity of radiation.

- (b) Given: Power of source = 6×10^{-2} Watts or Joules / s
 No of photons per second = 1×10^{16}

$$\text{Energy per photon} = \frac{\text{Power}}{\text{no of photons per second}}$$

$$= \frac{6 \times 10^{-2}}{1 \times 10^{16}}$$

$$E = hf = 6 \times 10^{-19} \text{ Joules}$$

$$\text{Energy} = hf - W_0 = eV_0 \text{ where } W_0 \text{ is work function.}$$

$$E - eV_0 = W_0 \text{ where } V_0 \text{ is stopping potential}$$

$$6 \times 10^{-19} - (1.6 \times 10^{-19}) \times (2.2) = W_0$$

$$6 \times 10^{-19} - 3.52 \times 10^{-19} = W_0$$

$$W_0 = 2.48 \times 10^{-19} \text{ Joules}$$

$$W_0 = \frac{2.48 \times 10^{-19}}{1.6 \times 10^{-19}} \text{ eV}$$

$$W_0 = 1.55 \text{ eV}$$

Photoelectric effect

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Zhe Xu



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 - The Rise of Digital Reading Physics Essay Answers Waec 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Physics Essay Answers Waec 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physics Essay Answers Waec 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physics Essay Answers Waec 2014
 - Personalized Recommendations
 - Physics Essay Answers Waec 2014 User Reviews and Ratings
 - Physics Essay Answers Waec 2014 and Bestseller Lists

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 - Physics Essay Answers Waec 2014 eBook Subscription Services
 - Physics Essay Answers Waec 2014 Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Physics Essay Answers Waec 2014 Compatibility with Devices
 - Physics Essay Answers Waec 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physics Essay Answers Waec 2014
 - Highlighting and Note-Taking Physics Essay Answers Waec 2014
 - Interactive Elements Physics Essay Answers Waec 2014
8. Staying Engaged with Physics Essay Answers Waec 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physics Essay Answers Waec 2014
9. Balancing eBooks and Physical Books Physics Essay Answers Waec 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physics Essay Answers Waec 2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physics Essay Answers Waec 2014
 - Setting Reading Goals Physics Essay Answers Waec 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physics Essay Answers Waec 2014
 - Fact-Checking eBook Content of Physics Essay Answers Waec 2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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